



JNTUH COLLEGE CODE: U2

EAMCET CODE: SDIP

SREE DATTHA INSTITUTE OF PHARMACY

(Approved by AICTE & PCI, New Delhi, Affiliated to JNTUH, Hyderabad, T.S)
Nagarjuna Sagar Road, Sheriguda (V), Ibrahimpatnam (M), R.R.Dist., Greater Hyderabad-501510. T.S
Ph.:+91-8801099936/35, 9393808082,
Email: principalsdip@sreedattha.ac.in, www.sreedattha.ac.in

2.5.3 Reforms in the process and procedure in the conduct of evaluation/examination; including Continuous Internal Assessment to improve the examination system.

Response:

B.Pharmacy

The JNTUH University, Hyderabad, to which the institute is affiliated is very active in introducing evaluation reforms to maintain high standards of quality. The institute has adopted the following reforms introduced by the university. Each under graduate programmes of 4 academic years (8 semesters) with the academic year being divided into two semesters each, each semester shall have - 'Continuous Internal Evaluation (CIE)' and Semester End Examination (SEE)'. Student have to attain two internal examination in each semester. Choice Based Credit System (CBCS) and Credit Based Semester 2 System (CBSS) as indicated by UGC and curriculum / course structure as suggested by AICTE are followed. Internal assessment as below:

Grading of weightage for both internal examinations.

The total internal evaluation carries 25% weightage for Mid Marks

1. 10 marks for Descriptive
2. 10 Marks for Objective
3. 5 Marks for Assignment.

M.Pharmacy

The M. Pharm Programmes in Pharmacy of JNTUH are of Semester pattern, with Four Semesters consisting of Two academic years, each academic year having Two Semesters. Each Semester shall have 'Continuous Internal Evaluation (CIE)' and 'SemesterEnd Examination (SEE)'. Choice Based Credit System (CBCS) and Credit Based Semester System (CBSS) are taken as 'references' for the present set of Regulations. The performance of a student in each semester shall be evaluated subject-wise (irrespective of credits assigned) for a maximum of 100 marks. The M. Pharm. project work (major project) will also be evaluated for 100 marks. For the theory subjects 75 marks shall be awarded for the performance in the Semester End Examination and 25 marks shall be awarded for Continuous Internal Evaluation (CIE).



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To.

The Coordinator,

NAAC, Bengaluru.

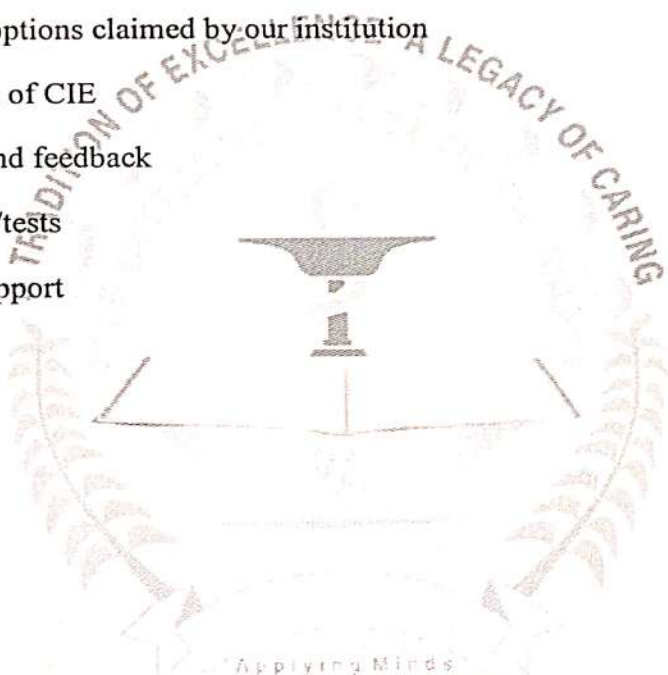
Subject: Policy document of the options claimed by the institution duly signed by the Head of the Institution

Reference: 2.5.4 The Institution provides opportunities to students for mid course improvement of performance through specific interventions.

Dear Sir/Madam,

Policy document of the options claimed by our institution

1. Timely administration of CIE
2. On time assessment and feedback
3. Make up assignments/tests
4. Remedial teaching/support



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11. Then respective subject teachers are directed to evaluate the answer sheets in stipulated time.
12. Evaluated Sessional theory answer papers are shown to the students.
13. Mark sheet of individual subject is displayed on notice board.
14. Any grievance regarding the evaluation like total of marks and/ or any discrepancy are sorted out during the showing of answer sheets itself.
15. Extra efforts should be plan for improvement of examination.
16. At the end of the year, the total internal marks (continuous assessment and Sessional exam) are shown to the students, which they sign upon confirmation of their marks.
17. Head of the institute supervises the entry of internal marks.



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Policy for Continuous Internal Evaluation

1. Each Semester will consist of a minimum of 15 weeks instructions. "ic. $15 \times 6 = 90$ instructional days (minimum instructional days)
2. 75% attendance for both theory and practical classes separately shall be mandatory to appear for Sessional examination and end semester examination.
3. Mid-term assessment will be of 50 marks which include 25 marks for theory and 25 marks practical Sessional
4. Two mid-term tests should be conducted.
5. Written test for 10 marks and Short quizzes for 10marks are selected for mid-term examination and 5 marks for internal assessment like regularity, promptness, performance.
6. There will be a minimum of one sessional examination of 25 marks conducted in each semester after minimum eight weeks of teaching from the date of commencement of that semester in each of the theory subjects during the semester.
7. For the practical head i) Major experimentii) minor experimentiii) Synopsis For the practical (Written test based on practical knowledge) iv) Daily assessment of practical for which includes day to day attendance, learners' skills (performance). viva, laboratory record, etc.
8. Examination schedule displayed at the start of each academic year. All the faculties are instructed to follow the schedule
9. Question paper for the mid-term examination should be submit by subject teachers well in advance
10. Students are kept aware about the timelines and duration by proper system. of examination by proper system.

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JNTUH COLLEGE CODE: U3

BAMCET CODE: BDIP

SREE DATTHA INSTITUTE OF PHARMACY

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Ph: +91-8801099936/35, 9393808082,
Email: principal@sdip@sreedattha.ac.in, www.sreedattha.ac.in

11. Solved answer paper of old question paper is given to the poor learners for practice.
12. Faculty should correct the mid-term answer sheet within three days of completion of exam and the answer sheet is shown to the students and allowing them for any doubt in correction.
13. Faculty should clarify the doubt of the students regarding correction.



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Email: principalsdip@sreedattha.ac.in, www.sreedattha.ac.in

POLICY DOCUMENT OF MIDCOURSE IMPROVEMENT OF PERFORMANCE OF STUDENTS

These following policies are followed by our institute for midcourse improvement of performance of our students

1. Two mid-term examinations are conducted in an academic year for B.Pharmacy and M. Pharmacy, three mid-term examinations are conducted for Pharm.D.
2. Mid-term examinations are conducted in time as per university academic calendar.
3. 50% of syllabus ie. 2 ½ unit has to be complete before the first mid-term examination remaining 50% before second mid-term examination.
4. 30% syllabus has to be complete before first mid-term examination for Pharm. D. 50% of remaining syllabus completed before 2nd and 3rd mid- term examination.
5. Every student has to submitted one assignment for each unit after completion of the unit.
6. Slip test is conducted for each unit after completion of the unit.
7. Assignment and slip test are corrected by the subject lecture and shown to the student
8. Lecture should give required suggestion for improvement in assignment and slip test.
9. Faculty should give the course notes in hand written format or print format for each unit.
10. Arrangement of remedial class for poor learner.

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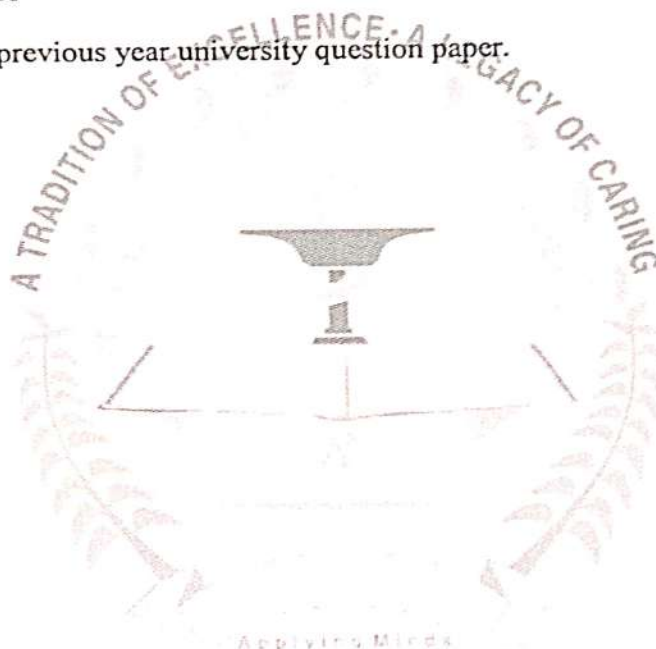
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List of opportunities provided for the students for midcourse improvement of performance in the examinations

1. Conduct of slip test
2. Conduct of unit test after completion of every unit
3. Assignment
4. Remedial class
5. Provide of course notes
6. Provide of answer of previous year university question paper.



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To,

The Coordinator,

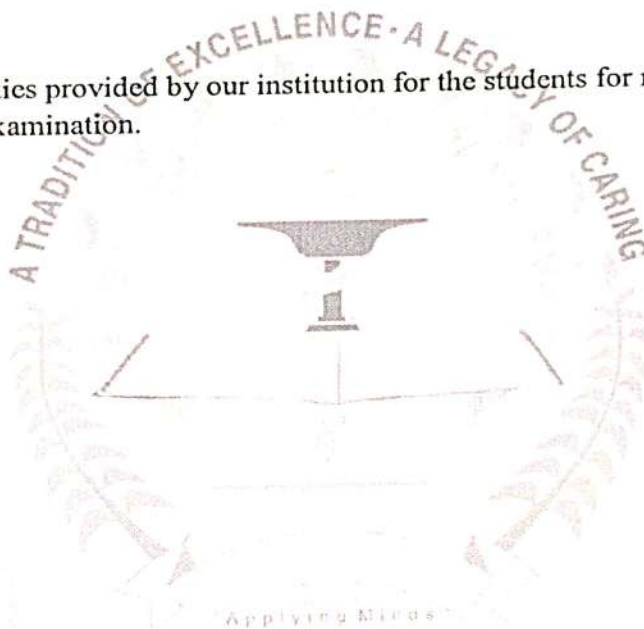
NAAC, Bengaluru.

Subject: List of opportunities provided for the students for midcourse improvement of performance in the examinations

Reference: 2.5.4 The Institution provides opportunities to students for midcourse improvement of performance through specific interventions.

Dear Sir/Madam,

These List of opportunities provided by our institution for the students for midcourse improvement of performance in the examination.



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Ph.:08414-202206, 320919, 9393808082

www.sreedattha.org, E-mail: principal@dip@sreedattha.ac.in

College Code: SDII

INVIGILATION DUTIES

PHARM D I and III Yr III MID EXAMINATION

Time: 10:30 AM - 12:30 PM

Duration of Exam: 2Hrs

S.No	DATE & DAY	ROOM NO	NAME OF THE INVIGILATOR
1.	15-07-2024 MONDAY	1	Dr. K. Lakshmi Deepthi
		2	Mrs. K. Priyanka
2.	16-07-2024 TUESDAY	1	Mrs. J. Prathyusha
		2	Dr. Y. Ramulu
3.	18-07-2024 THURSDAY	1	Mr. D. Vishwanath
		2	Dr. Amina Turki
4.	19-07-2024 FRIDAY	1	Mrs. V. Jhansi Laxmi
		2	Mrs. G. Deepthi
5.	20-07-2024 SATURDAY	1	Mrs. G. Swathi
		2	Dr. Amina Turki
6.	22-07-2024 MONDAY	1	Dr. K. Lakshmi Deepthi
		2	Mrs. K. Priyanka

Principal
Principal

Sree Dattha Institute Of Pharmacy
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College Code: SDIP

INVIGILATION DUTIES FOR B PHARMACY & M. PHARMACY END EXAMINATION

Exam Timings: Forenoon Session: 10.00 AM - 01.00 PM

Afternoon Session: 02.00 PM - 05.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME	
		FORENOON SESSION	AFTERNOON SESSION
28-03-2024 THURSDAY	1	Mrs. K. Priyanka	Mr. V. Kalyan Varma (2.00 to 3.30) Mr. S. Ramakrishna (3.30 to 5.00)
	2	-----	Mrs. V. Jhansi Laxmi (2.00 to 3.30) Dr. Y. Ganesh Kumar (3.30 to 5.00)
	3	-----	Mr. D. Vishwanath
	4	-----	Mrs. M. Jyothi (2.00 to 3.30) Dr. Y. Ramulu (3.30 to 5.00)
	5	-----	Mrs. K. Lakshmi Deepthi (2.00 to 3.30) Mr. B. Ramakrishna (3.30 to 5.00)





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College Code: SDIP

INVIGILATION DUTIES FOR B PHARMACY & M. PHARMACY END EXAMINATION

Exam Timings: Forenoon Session: 10.00 AM - 01.00 PM

Afternoon Session: 02.00 PM - 05.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME	
		FORENOON SESSION	AFTERNOON SESSION
22-03-2024 FRIDAY	1	Mr. V. Kalyan Varma	Dr. Y. Ganesha Kumar
	2	Ms. Sayeda Hina Naaz	Dr. Y. Ramulu
	3	Mrs. J. Prathyusha	Mr. D. Vishwanath
	4	Mrs. K. Lakshmi Deepthi	Mrs. V. Jhansi Laxmi (2.00 to 3.30) Mr. B. Ramakrishna (3.30 to 5.00)
	5	Mrs. M. Jyothi	-----
	6	Mrs. G. Swathi	-----
	7	Mrs. G. Deepthi	-----

UPADHAR
Principal
Sree Dattha Institute Of Pharmacy
Sheriguda(V), Ibrahimpatnam(M),
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INVIGILATION DUTIES FOR B PHARMACY I YR END EXAMINATION

Exam Timings: Forenoon Session: 10.00 AM - 01.00 PM

Afternoon Session: 02.00 PM - 05.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME
		AFTERNOON SESSION
16-03-2024 SATURDAY	1	Mr. B. Ramakrishna
	2	Dr. Y. Ramulu
	3	Mrs. M. jyothi
	4	Mrs. G. Swathi (2.00 to 3.30) Dr. Y. Ganesh Kumar(3.30 to 5.00)
	5	Ms. Shashirekha
	6	Mrs. A. Niharika (2.00 to 3.30) Mr. R. Suresh (3.30 to 5.00)
	7	Ms. Sayeda Hina Naaz (2.00 to 3.30) Mr. S. Ramakrishna (3.30 to 5.00)

UPadela
Principal
Sree Dattha Institute Of Pharmacy
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College Code: SDIP

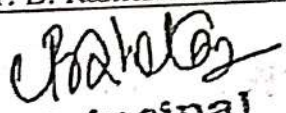
INVIGILATION DUTIES FOR B PHARMACY & M PHARMACY SEM EXAMINATION

Exam Timings:

Forenoon Session: 10.00 AM - 01.00 PM

Afternoon Session: 02.00 PM - 05.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME	
		FORENOON SESSION	AFTERNOON SESSION
15-03-2024 FRIDAY	1	Mrs. K. Priyanka	Mr. V. Kalyan Varma (2.00 to 3.30) Ms. Shashirekha (3.30 to 5.00)
	2	Mrs. G. Deepthi	Mr. D. Vishwanath
	3	-----	Mrs. J. Prathyusha (2.00 to 3.30) Mr. B. Ramakrishna (3.30 to 5.00)


Principal
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INVIGILATION DUTIES FOR B PHARMACY I YR & III YR END EXAMINATION

Exam Timings:

Forenoon Session: 10.00 AM - 01.00 PM

Afternoon Session: 02.00 PM - 05.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME	
		FORENOON SESSION	AFTERNOON SESSION
14-03-2024 THURSDAY	1	Mrs. G. Swathi	Mr. B. Ramakrishna
	2	Mrs. V. Jhansi Laxmi	Dr. Y. Ganesh Kumar
	3	Mrs. P. Rajitha	Dr. Y. Ramulu
	4	Mrs. M. Jyothi	Mr. V. Kalyan Varma (2.00 to 3.30) Mr. R Sukesh (3.30 to 5.00)
	5	Mrs. A. Niharika	Ms. Shashirekha
	6	Dr. Amatul Ali Sameera	Mrs. K. Lakshmi Deepthi (2.00 to 3.30) Mr. S. Ramakrishna (3.30 to 5.00)
	7	-----	Ms. Sayeda Hina Naaz (2.00 to 3.30) Mr. M. Anand (3.30 to 5.00)

Principal

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INVIGILATION DUTIES FOR B PHARMACY I YR & III YR END EXAMINATION

Exam Timings: Forenoon Session: 10.00 AM - 01.00 PM

Afternoon Session: 02.00 PM - 05.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME	
		FORENOON SESSION	AFTERNOON SESSION
12-03-2024 TUESDAY	1	Mrs. G. Swathi	Mr. B. Ramakrishna
	2	Mrs. V. Jhansi Laxmi	Dr. Y. Ganesha Kumar
	3	CH.S. Vijaya Vani	Dr. Y. Ramulu
	4	Mrs. M. Jyothi	Mr. D. Vishwanath
	5	Mrs. G. Deepthi	Mrs. A. Niharika (2.00 to 3.30) Ms. Shashirekha (3.30 to 5.00)
	6	Mr. V. Kalyan Varma	Dr. Amatul Ali Sameera (2.00 to 3.30) Mr. S. Ramakrishna (3.30 to 5.00)
	7	-----	Mrs. K. Lakshmi Deepthi (2.00 to 3.30) Mr. S. Ramakrishna (3.30 to 5.00)

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Exam Timings: Forenoon Session: 10.00 AM - 01.00 PM

Afternoon Session: 02.00 PM - 05.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME	
		FORENOON SESSION	AFTERNOON SESSION
09-03-2024 SATURDAY	1	Mrs. A. Niharika	Mr. B. Ramakrishna
	2	Dr. Amatul Ali Sameera (10.00 to 11.30) Ms. Shashirekha (11.30 to 1.00)	Dr. Y. Ganesh Kumar
	3	CH.S. Vijaya Vani	Dr. Y. Ramulu
	4	Mrs. K. Priyanka	Mrs. Rupali Gawande
	5	Ms. Syed Hina Naaz	Mrs. J. Prathyusha
	6	Dr. Amina (10.00 to 11.30) Mrs. P. Rajitha (11.30 to 1.00)	Mrs. M. Jyothi
	7	-----	Mr. S. Ramakrishna

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College Code: SI

INVIGILATION DUTIES FOR B PHARMACY III & IV YEAR I SEM I & II MID EXAMINATION

B. Pharmacy Exam Timings: Forenoon Session: 10.00 AM - 11.30 AM

Afternoon Session: 02.00 PM - 03.30 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME		
		FORENOON SESSION	ROOM NO	AFTERNOON SESSION
13-12-2023 WEDNESDAY	404	Mrs. A. Niharika	404	Mr. D. Vishwanath
	405	Mrs. Rupali Gawande	405	Mrs. J. Prathyusha
	406	Mrs. Laxmi Deepthi	406	Dr. G. Shalini
	407	Dr. Y. Ramulu	407	Dr. Amatul Ali Sameera
14-12-2023 THURSDAY	404	Mrs. G. Swathi	404	Mrs. A. Niharika
	405	Mrs. G. Deepthi	405	Dr. Y. Ramulu
	406	Mr. B. Ramakrishna	406	Dr. Amina
	407	Mrs. M. Jyothi	407	Mr. D. Vishwanath
15-12-2023 FRIDAY	404	Mrs. J. Prathyusha	---	---
	405	Mrs. Rupali Gawande	---	---
	406	Mr. B. Ramakrishna	---	---
	407	Mrs. Laxmi Deepthi	---	---



Principal
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College Code: SDIF

INVIGILATION DUTIES FOR B PHARMACY I & II YEAR II SEM II MID EXAMINATION

B. Pharmacy Exam Timings: Forenoon Session: 10.00 AM - 11.30 AM
Afternoon Session: 02.00 PM - 03.30 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME		
		FORENOON SESSION	ROOM NO	AFTERNOON SESSION
11-09-2023 MONDAY	109	Mrs. G. Swathi	109	Mr. D. Vishwanath
	205	Dr. S. Vijaya Vani	205	Mrs. Maimunna Khatun
	206	Mrs. V. Jhansi Laxmi	206	Dr. Y. Ganesh Kumar
	305	Dr. Sumera Iram	305	Mr. E. Nikhil Charvarthy
	306	Mr. Taqiuddin	306	Mrs. A. Niharika
12-09-2023 TUESDAY	109	Mrs. CH. Madhavi	109	Mrs. G. Swathi
	205	Mrs. J. Prathyusha	205	Mr. Taqiuddin
	206	Mrs. Maimunna Khatun	206	Dr. Sumera Iram
	305	Mrs. Rupali Gawande	305	Mr. Kalyan Varma
	306	Mrs. G. Deepthi	306	Mr. E. Nikhil Charvarthy
13-09-2023 WEDNESDAY	109	Mrs. V. Jhansi Laxmi	----	-----
	205	Mrs. CH. Madhavi	----	-----
	206	Dr. Y. Ganesh Kumar	----	-----
	305	Mrs. A. Niharika	----	-----
	306	Mrs. Rupali Gawande	----	-----



DR. S A SREENIVAS
M. Pharm, Ph. D

PRINCIPAL
Sree Dattha Institute of Pharmacy
Sheriguda, V Ibrahimpatnam, R.R. Dist.-501510.



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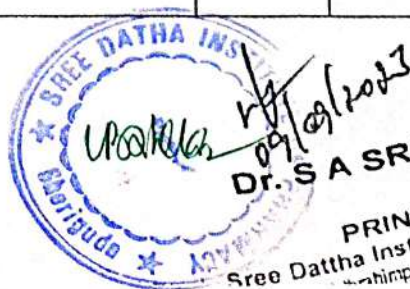
College Code: SI

INVIGILATION DUTIES FOR B PHARMACY I & II YEAR II SEM II MID EXAMINATION

B. Pharmacy Exam Timings: Forenoon Session: 10.00 AM - 11.30 AM

Afternoon Session: 02.00 PM - 03.30 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME		
		FORENOON SESSION	ROOM NO	AFTERNOON SESSION
11-09-2023 MONDAY	109	Mrs. G. Swathi	109	Mr. D. Vishwanath
	205	Dr. S. Vijaya Vani	205	Mrs. Maimunna Khatun
	206	Mrs. V. Jhansi Laxmi	206	Dr. Y. Ganesh Kumar
	305	Dr. Sumera Iram	305	Mr. E. Nikhil Charvarthy
	306	Mrs. A. Niharika	306	Dr. Amatul Ali Sameera
12-09-2023 TUESDAY	109	Mrs. CH. Madhavi	109	Mrs. G. Swathi
	205	Mrs. J. Prathyusha	205	Mr. Taquiuddin
	206	Mrs. Maimunna Khatun	206	Dr. Sumera Iram
	305	Mrs. Rupali Gawande	305	Mr. Kalyan Varma
	306	Mrs. G. Deepthi	306	Mr. E. Nikhil Charvarthy
13-09-2023 WEDNESDAY	109	Mrs. V. Jhansi Laxmi	----	-----
	205	Mrs. CH. Madhavi	----	-----
	206	Mr. Taquiuddin	----	-----
	305	Mrs. A. Niharika	----	-----
	306	Mrs. Rupali Gawande	----	-----



DR. S A SREENIVAS
M. Pharm, Ph. D
PRINCIPAL
Sree Dattha Institute of Pharmacy
Sheriguda, Ibrahimpatnam, R.R. Dist.-501510



SREE DATTHA INSTITUTE OF PHARMACY

(Approved by AICTE & PCI, New Delhi, Affiliated to JNTU, Hyderabad)

Sagar Road, Sheriguda, Ibrahimpatnam, R.R. Dist.-501510

Ph.:08414-202206, 320919, 9393808082

www.sreedattha.org, E-mail: principalsdip@sreedattha.ac.in

College Code: SDIP

INVIGILATION DUTIES FOR PHARM D and M PHARMACY SEM END EXAMINATION

Exam Timings: Forenoon Session: 10.00 AM - 01.00 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME
		FORENOON SESSION
04-09-2023 MONDAY	1	Mrs. Maimunna Khathun
	2	Mr. B. Ramakrishna <i>BB</i>
05-09-2023 TUESDAY	1	Mr. V. Kalyan Varma <i>V</i>
06-09-2023 WEDNESDAY	1	Mr. Y. Ramulu
	2	Mrs. R. Prathima Mrs. H. Jyothi
	3	Mr. U. Sambamoorthy
08-09-2023 FRIDAY	1	Dr. Sumera Iram <i>Sumera</i>
	2	Mr. E. Nikhil Chakravarthy <i>E.N.P.</i>
09-09-2023 SATURDAY	1	Mrs. A. Niharika <i>Niharika</i>
11-09-2023 MONDAY	1	Mr. V. Kalyan Varma <i>V</i>
	2	Mr. Y. Ramulu
13-09-2023 WEDNESDAY	1	Mr. E. Nikhil Chakravarthy Mrs. J. Prathima
	2	Mr. D. Vishwanath (10.00 to 11.30) Dr. Amatul Ali Sameera (11.30 to 01.00)
15-09-2023 FRIDAY	1	Mrs. G. Deepthi
	2	Mr. B. Ramakrishna <i>BB</i>

Dr. S A SREENIVAS
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College Code: SI

INVIGILATION DUTIES FOR B PHARMACY III YEAR II SEM II MID EXAMINATION

B. Pharmacy Exam Timings: Forenoon Session: 10.00 AM - 11.30 AM

Afternoon Session: 02.00 PM - 03.30 PM

DATE & DAY	ROOM NO	INVIGILATOR NAME		
		FORENOON SESSION	ROOM NO	AFTERNOON SESSION
26-06-2023 MONDAY	404	Mr. U. Samba Moorthy	404	Dr. Amatul Ali Sameera
	405	Dr. Y. Ganesh Kumar	405	Mr. B. Ramakrishna
	406	Mrs. Maimunna Khatun	406	Mr. E. Nikhil Chakravarthy
	407	Ms. G. Deepthi	407	Mr. Kalyan Varma
27-06-2023 TUESDAY	404	Mrs. G. Mary Ratna Anitha	404	Mr. B. Ramakrishna
	405	Mr. U. Samba Moorthy	405	Dr. Amatul Ali Sameera
	406	Mr. Kalyan Varma	406	Dr. Sumera Iram
	407	Mr. E. Nikhil Chakravarthy	407	Mr. Kalyan Varma
28-06-2023 WEDNESDAY	404	Dr. S. Vijaya Vani	----	-----
	405	Mr. D. Vishwanath	----	-----
	406	Mrs. Maimunna Khatun	----	-----
	407	Dr. Y. Ganesh Kumar	----	-----

Dr. S A SREENIVAS
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College Code: SDIP

INVIGILATION DUTIES

PHARM D III Yr III MID EXAMINATION

Pharm.D Exam Timings: Forenoon Session: 10.00 AM - 12.00 AM

Afternoon Session: 01.30 PM - 03.30 PM

Duration of Exam: 2Hrs

INVIGILATOR NAME				
DATE & DAY	ROOM NO	FORENOON SESSION	ROOM NO	AFTERNOON SESSION
26-06-2023 MONDAY	109	Mr. D. Vishwanath	109	Dr. Sumeera Iram
27-06-2023 TUESDAY	109	Mrs. G. Deepthi	109	Ms. Shashi Rekha
28-06-2023 WEDNESDAY	109	Dr. Amatul Ali Sameera	109	Dr. Sumeera Iram

Dr. S A SREENIVAS
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OFFICE OF THE CONTROLLER OF EXAMINATIONS AND ADMISSIONS
SHRI JAGDISH PRASAD JHABARMAL TIBREWALA UNIVERSITY,
VIDYANAGARI, JHUNJHUNU

NOTIFICATION

DATED: 04/12/2014

Whereas the reports received from the examiners on the Ph.D. thesis and Viva Voce of the following candidates, the Vice-Chancellor, on behalf of the Academic Council, has approved award of Ph.D. degree to the candidate concerned for the year 2014 in accordance with the Act of JJTU.

S.No	Enrollment No.	Name of the candidate/Viva Voce	Subject/Faculty	Topic	Name of the Supervisor
1	PEN/TH/11/00169	Nawale Roshan Murlidhar 26/11/2014	Pharmacy (Faculty of Pharmacy)	STIMULI SENSITIVE POLYMERIC DRUG DELIVERY SYSTEM	Dr.Rahul Mayee (JJT/2K9/SC/0589)
2	PEN/PH/10/00275	Ravikiran Balajirao Wakade 26/11/2014	Pharmacy (Faculty of Pharmacy)	DESIGN AND CHARACTERIZATION OF FLOATING DRUG DELIVERY SYSTEMS AND STUDY ON FACTORS CONTROLLING THEIR BUOYANCY AND RELEASE PATTERN	Dr.S.A Sreenivas (JJT/2K9/SC/0323)

The date of Viva Voce shall be the date of award of the degree as mentioned above the names of the concerned candidates.

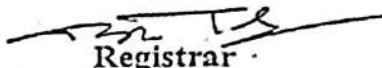
JJT/Ph.D./Exam/278

26/11/2014

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Academic calander

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

PHARM. D (REGULAR) AND PHARM. D (PB) II YEAR

Pharm. D (Regular) and Pharm. D (PB) II Year

S. No	Description	Duration	
		From	To
1	Commencement of classwork	01.11.2023	
2	1 st Spell of Instructions	01.11.2023	30.01.2024 (13 Weeks)
3	First Mid Term Examinations	31.01.2024	05.02.2024 (1 Week)
4	Submission of First Mid Term Exam Marks to the University on or before	09.02.2024	
5	2 nd Spell of Instructions	06.02.2024	01.05.2024 (12 Weeks)
6	Supplementary Examinations	18.03.2024	23.03.2024 (1 Week)
7	Second Mid Term Examinations	02.05.2024	07.05.2024 (1 Week)
8	Submission of Second Mid Term Exam Marks to the University on or before	10.05.2024	
9	3 rd Spell of Instructions (including summer vacation)	08.05.2024	13.08.2024 (14 Weeks)
10	Summer vacation	13.05.2024	25.05.2024 (2 Weeks)
11	Third Mid Term Examinations	14.08.2024	20.08.2024 (1 Week)
12	Preparation Holidays and Practical Examinations	21.08.2024	03.09.2024 (2 Weeks)
13	Submission of Third Mid Term Exam Marks to the University on or before	28.08.2024	
14	End / Supplementary Examinations	04.09.2024	20.09.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 200

S. Tan Kalyan
REGISTRAR I/c.



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD
ACADEMIC CALENDAR 2023-24
M. Pharm, II YEAR I & II SEMESTERS

I SEM

S. No	Description	Duration	
		From	To
1	Commencement of I Semester classwork		11.09.2023
2	1 st Spell of Instructions (including Dussehra Recess)	11.09.2023	18.11.2023 (10 Weeks)
3	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	Preparation of Dissertation Work Proposals	11.09.2023	07.10.2023 (4 Weeks)
5	Dissertation Work Review-I: (Dissertation Submission & approval)	09.10.2023	13.10.2023 (1 Week)
6	Last date for submission of list of approved DRC - I students from the College to the University Examination branch.		18.10.2023
7	First Mid Term Examinations	20.11.2023	25.11.2023 (1 Week)
8	Submission of First Mid Term Exam Marks to the University on or before		30.11.2023
9	2 nd Spell of Instructions	28.11.2023	02.02.2024 (9 Weeks)
10	Second Mid Term Examinations	03.02.2024	09.02.2024 (1 Week)
11	Comprehensive Viva Voce	12.02.2024	17.02.2024 (1 Week)
12	Dissertation Work Review - II (Phase - I)	18.02.2024	23.02.2024 (1 Week)
13	Submission of Second Mid Term Exam Marks to the University on or before		15.02.2024
14	Last date for submission of Dissertation Work Review - II (Phase - I) Marks & Comprehensive Viva Voce marks to the University on or before		24.02.2024
15	End Semester Examinations	26.02.2024	07.03.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 100

II SEM

S. No	Description	Duration	
		From	To
1	Commencement of II Semester (Dissertation Work Continuation) (16.10.2023 to 07.03.2024 - 20 weeks)		11.03.2024
2	** Dissertation Work Review - II (Phase-II)	11.03.2024	16.03.2024 (1 Week)
3	Last date for submission of Dissertation Work Review - II (Phase-II) Marks to the University on or before		22.03.2024
4	Dissertation Work Review - III (Phase - I) (18.03.2024 to 27.07.2024 - 19 Weeks)	29.07.2024	03.08.2024 (1 Week)
5	Last date for submission of Dissertation Work Review-III (Phase-I) Marks to the University on or before		09.08.2024
6	* Date of Eligibility of Thesis Submission		09.08.2024
7	Submission of Thesis and Dissertation Viva -Voce Examination (DRC -III Phase-I)		--
8	** Dissertation Work Review - III (Phase -II) (05.08.2024 to 04.11.2024 - 13 Weeks)	05.11.2024	09.11.2024 (1 Week)
9	Last date for submission of Dissertation Work Review -III (Phase-II) Marks		11.11.2024
10	Submission of Thesis and Dissertation Viva -Voce Examination (Phase-II) follows		--

* After completion of 40 weeks from the date of approval of Dissertation work proposal and subject to approval of Dissertation Work Review-III.

** Phase-II will be conducted only for unsuccessful students in Phase -I

Note: The unsuccessful students in Dissertation Work Review-II (Phase-II) shall appear for Dissertation Work Review-III at the time of Dissertation Work Review-III. These students shall reappear for Dissertation Work Review-III in the next academic year at the time of Dissertation Work Review -I only after completion of Dissertation Work Review -II, and then Dissertation Work Review -III follows.

The Dissertation Viva-Voce External examination Marks must be submitted on the day of examination to the University.



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21/11/24

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

M. Pharm. I YEAR I & II SEMESTERS

I SEM

S. No	Description	Duration	
		From	To
1	Commencement of I Semester classwork	09.10.2023	
2	1 st Spell of Instructions (including Dussehra Recess)	09.10.2023	16.12.2023 (10 Weeks)
3	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	First Mid Term Examinations	18.12.2023	23.12.2023 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	29.12.2023	
6	2 nd Spell of Instructions	27.12.2023	29.02.2024 (9 Weeks)
7	Second Mid Term Examinations	01.03.2024	06.03.2024 (1 Week)
8	Preparation Holidays and Practical Examinations	07.03.2024	14.03.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	13.03.2024	
10	End Semester Examinations	15.03.2024	30.03.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 100

II SEM

S. No	Description	Duration	
		From	To
1	Commencement of II Semester classwork	01.04.2024	
2	1 st Spell of Instructions	02.04.2024	22.06.2024 (11 Weeks)
3	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
4	First Mid Term Examinations	24.06.2024	29.06.2024 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	03.07.2024	
6	2 nd Spell of Instructions	01.07.2024	31.08.2024 (9 Weeks)
7	Second Mid Term Examinations	02.09.2024	06.09.2024 (1 Week)
8	Preparation Holidays and Practical Examinations	09.09.2024	13.09.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	11.09.2024	
10	End Semester Examinations	17.09.2024	28.09.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 100

[Signature]
29/9/23
REGISTRAR



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

PHARM. D (REGULAR) AND PHARM. D (PB) I YEAR

Pharm. D (Regular) and Pharm. D (PB) I Year

S. No	Description	Duration	
		From	To
1	Commencement of classwork / Orientation programme	29.09.2023	
2	1 st Spell of Instructions (including Dussehra Recess)	29.09.2023	30.12.2023 (14 Weeks)
3	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	First Mid Term Examinations	02.01.2024	06.01.2024 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	12.01.2024	
6	2 nd Spell of Instructions	08.01.2024	30.03.2024 (12 Weeks)
7	Supplementary Examinations	18.03.2024	23.03.2024 (1 Week)
8	Second Mid Term Examinations	01.04.2024	06.04.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	12.04.2024	
10	3 rd Spell of Instructions (including Summer Vacation)	08.04.2024	16.07.2024 (14 Weeks)
11	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
12	Third Mid Term Examinations	18.07.2024	24.07.2024 (1 Week)
13	Preparation Holidays and Practical Examinations	25.07.2024	06.08.2024 (2 Weeks)
14	Submission of Third Mid Term Exam Marks to the University on or before	30.07.2024	
15	End / Supplementary Examinations	07.08.2024	22.08.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 202


14/9/23
REGISTRAR



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

B. Pharm II YEAR I & II SEMESTERS

I SEM

S. No	Description	Duration	
		From	To
1	Commencement of I Semester classwork	09.10.2023	
2	1 st Spell of Instructions (including Dussehra Recess	09.10.2023	16.12.2023 (10 Weeks)
	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
3	First Mid Term Examinations	18.12.2023	23.12.2023 (1 Week)
4	Submission of First Mid Term Exam Marks to the University on or before	30.12.2023	
5	2 nd Spell of Instructions	27.12.2023	29.02.2024 (9 Weeks)
6	Second Mid Term Examinations	01.03.2024	07.03.2024 (1 Week)
7	Preparation Holidays and Practical Examinations	11.03.2024	16.03.2024 (1 Week)
8	Submission of Second Mid Term Exam Marks to the University on or before	13.03.2024	
9	End Semester Examinations	18.03.2024	30.03.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 101

II SEM

S. No	Description	Duration	
		From	To
1	Commencement of II Semester classwork	02.04.2024	
2	1 st Spell of Instructions (including Summer Vacation)	02.04.2024	22.06.2024 (12 Weeks)
3	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
4	First Mid Term Examinations	24.06.2024	29.06.2024 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	06.07.2024	
6	2 nd Spell of Instructions	01.07.2024	31.08.2024 (9 Weeks)
7	Second Mid Term Examinations	02.09.2024	06.09.2024 (1 Week)
8	Preparation Holidays and Practical Examinations	09.09.2024	13.09.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	11.09.2024	
10	End Semester Examinations	17.09.2024	30.09.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 100



S. Tan Kaly
REGISTRAR I/c.

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

B. Tech./B. Pharm. III YEAR I & II SEMESTERS

I SEM

S. No	Description	Duration	
		From	To
1	Commencement of I Semester classwork	11.10.2023	
2	1 st Spell of Instructions (including Dussehra Recess)	11.10.2023	12.12.2023 (9 Weeks)
3	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	First Mid Term Examinations	13.12.2023	19.12.2023 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	23.12.2023	
6	2 nd Spell of Instructions	20.12.2023	19.02.2024 (8 Weeks)
7	Second Mid Term Examinations	20.02.2024	24.02.2024 (1 Week)
8	Preparation Holidays and Practical Examinations	26.02.2024	02.03.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	28.02.2024	
10	End Semester Examinations	04.03.2024	16.03.2024 (2 Weeks)

Note: No. of Working/ instructional days: 90

II SEM

S. No	Description	Duration	
		From	To
1	Commencement of II Semester classwork	18.03.2024	
2	1 st Spell of Instructions	18.03.2024	10.05.2024 (8 Weeks)
3	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
4	First Mid Term Examinations	27.05.2024	01.06.2024 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	06.06.2024	
6	2 nd Spell of Instructions (including Summer Vacation)	03.06.2024	03.08.2024 (9 Weeks)
7	Second Mid Term Examinations	05.08.2024	09.08.2024 (1 Week)
8	Preparation Holidays and Practical Examinations	12.08.2024	17.08.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	14.08.2024	
10	End Semester Examinations	19.08.2024	31.08.2024 (2 Weeks)

Note: No. of Working/ instructional days: 90



S. Tan Kaly
REGISTRAR I/c.

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

Pharm. D (Regular) and Pharm. D (PB) III Year

Pharm. D (Regular) III Year

S. No	Description	Duration	
		From	To
1	Commencement of classwork		19.09.2023
2	1 st Spell of Instructions (including Dussehra Recess)	19.09.2023	23.12.2023 (13 Weeks)
3	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	First Mid Term Examinations	27.12.2023	03.01.2024 (1 Week)
5	2 nd Spell of Instructions	04.01.2024	30.03.2024 (12 Weeks)
6	Submission of First Mid Term Exam Marks to the University on or before		09.01.2024
7	Supplementary Examinations	05.02.2024	12.02.2024
8	Second Mid Term Examinations	01.04.2024	06.04.2024 (1 Week)
9	3 rd Spell of Instructions	08.04.2024	12.07.2024 (14 Weeks)
10	Submission of Second Mid Term Exam Marks to the University on or before		12.04.2024
11	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
12	Third Mid Term Examinations	15.07.2024	20.07.2024 (1 Week)
13	Preparation Holidays and Practical Examinations	22.07.2024	03.08.2024 (2 Weeks)
14	Submission of Third Mid Term Exam Marks to the University on or before		27.07.2024
15	End / Supplementary Examinations	05.08.2024	17.08.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 202

Pharm. D (PB) III Year

S. No	Description	Duration	
		From	To
1	Commencement of internship in general ward	19.09.2023	18.03.2024 (6 Months)
2	Report submission of internship in general ward		19.03.2024
3	Commencement of internship in Specialty ward -1	20.03.2024	18.05.2024 Months)
4	Report submission of internship in Specialty ward -1		20.05.2024
5	Commencement of internship in Specialty ward -2	21.05.2024	20.07.2024 Months)
6	Report submission of internship in Specialty ward-2		22.07.2024
7	Commencement of internship in Specialty ward -3	23.07.2024	21.09.2024 (2 Months)
8	Report submission of internship in Specialty ward -3		23.09.2024
9	Final viva of Internship		30.09.2024



REGISTRAR

12/9/23

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

B. Pharm. I YEAR I & II SEMESTERS

I SEM

S. No	Description	Duration	
		From	To
1	Commencement of I Semester classwork / Orientation programme	29.09.2023	
2	1 st Spell of Instructions (including Induction programme)	29.09.2023	08.12.2023 (10 Weeks)
3	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	First Mid Term Examinations	11.12.2023	16.12.2023 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	21.12.2023	
6	2 nd Spell of Instructions	18.12.2023	17.02.2024 (9 Weeks)
7	Second Mid Term Examinations	19.02.2024	24.02.2024 (1 Week)
8	Preparation Holidays and Practical Examinations	26.02.2024	02.03.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	28.02.2024	
10	End Semester Examinations	04.03.2024	19.03.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 101

II SEM

S. No	Description	Duration	
		From	To
1	Commencement of II Semester classwork	21.03.2024	
2	1 st Spell of Instructions (including Summer Vacation)	21.03.2024	07.06.2024 (11 Weeks)
3	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
4	First Mid Term Examinations	10.06.2024	15.06.2024 (1 Week)
5	Submission of First Mid Term Exam Marks to the University on or before	21.06.2024	
6	2 nd Spell of Instructions	18.06.2024	22.08.2024 (9 Weeks)
7	Second Mid Term Examinations	23.08.2024	30.08.2024 (1 Week)
8	Preparation Holidays and Practical Examinations	31.08.2024	06.09.2024 (1 Week)
9	Submission of Second Mid Term Exam Marks to the University on or before	03.09.2024	
10	End Semester Examinations	09.09.2024	21.09.2024 (2 Weeks)

Note: No. of Working / Instructional Days: 100


 14/9/23
 REGISTRAR



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

Revised ACADEMIC CALENDAR 2023-24

Pharm D. (Regular) IV Year

Pharm. D. (Regular) IV Year

S. No	Description	Duration	
		From	To
1	Commencement of classwork	25.07.2023	
2	1 st Spell of Instructions	25.07.2023	13.10.2023 (12 Weeks)
3	First Mid Term Examinations	16.10.2023	21.10.2023 (1 Week)
	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	2 nd Spell of Instructions	30.10.2023	27.01.2024 (12 Weeks)
5	Submission of First Mid Term Exam Marks to the University on or before	04.11.2023	
6	Supplementary Examinations	06.11.2023	13.11.2023
7	Second Mid Term Examinations	29.01.2024	02.02.2024 (1 Week)
8	3 rd Spell of Instructions	03.02.2024	04.05.2024 (13 Weeks)
9	Submission of Second Mid Term Exam Marks to the University on or before	09.02.2024	
10	Third Mid Term Examinations	06.05.2024	11.05.2024 (1 Week)
11	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
12	Submission of Third Mid Term Exam Marks to the University on or before	27.05.2024	
13	Preparation Holidays and Practical Examinations	27.05.2024	01.06.2024 (1 Week)
14	End / Supplementary Examinations	03.06.2024	15.06.2024 (2 Weeks)

24/7/23
REGISTRAR



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

Pharm D. (Regular) V & VI Year

Pharm. D. (Regular) V Year

S. No	Description	Duration	
		From	To
1	Commencement of classwork	29.05.2023	
2	1 st Spell of Instructions	29.05.2023	19.08.2023 (12 Weeks)
3	First Mid Term Examinations	21.08.2023	26.08.2023 (1 Week)
4	2 nd Spell of Instructions (including Dussehra Recess)	28.08.2023	25.11.2023 (13 Weeks)
5	Submission of First Mid Term Exam Marks to the University on or before	02.09.2023	
6	Supplementary Examinations	04.09.2023	09.09.2023 (1 Week)
7	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
8	Second Mid Term Examinations	28.11.2023	02.12.2023 (1 Week)
9	3 rd Spell of Instructions	04.12.2023	24.02.2024 (12 Weeks)
10	Submission of Second Mid Term Exam Marks to the University on or before	08.12.2023	
11	Third Mid Term Examinations	26.02.2024	02.03.2024 (1 Week)
12	Preparation Holidays and Practical Examinations	04.03.2024	16.03.2024 (2 Weeks)
13	Submission of Third Mid Term Exam Marks to the University on or before	16.03.2024	
14	End / Supplementary Examinations	18.03.2024	30.03.2024 (2 Weeks)

Pharm. D (Regular) VI Year

S. No	Description	Duration	
		From	To
1	Commencement of internship in General ward	29.05.2023	28.11.2023 (6 Months)
2	Report submission of internship in General ward	29.11.2023	
3	Commencement of internship in Specialty ward -1	30.11.2023	29.01.2024 (2 Months)
4	Report submission of internship in Specialty ward -1	30.01.2024	
5	Commencement of internship in Specialty ward - 2	31.01.2024	29.03.2024 (2 Months)
6	Report submission of internship in Specialty ward-2	30.03.2024	
7	Commencement of internship in Specialty ward - 3	01.04.2024	31.05.2024 (2 Months)
8	Report submission of internship in Specialty ward - 3	01.06.2024	
9	Final viva of internship	04.06.2024	



[Signature]
31/5/23
REGISTRAR

JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD

ACADEMIC CALENDAR 2023-24

Pharm D. (Regular) IV Year

Pharm. D. (Regular) IV Year

S. No	Description	Duration	
		From	To
1	Commencement of classwork	24.07.2023	
2	1 st Spell of Instructions	24.07.2023	13.10.2023 (12 Weeks)
3	First Mid Term Examinations	16.10.2023	21.10.2023 (1 Week)
	Dussehra Recess	23.10.2023	28.10.2023 (1 Week)
4	2 nd Spell of Instructions	30.10.2023	25.01.2024 (12 Weeks)
5	Submission of First Mid Term Exam Marks to the University on or before	04.11.2023	
6	Supplementary Examinations	06.11.2023	13.11.2023
7	Second Mid Term Examinations	27.01.2024	02.02.2024 (1 Week)
8	3 rd Spell of Instructions	03.02.2024	04.05.2024 (13 Weeks)
9	Submission of Second Mid Term Exam Marks to the University on or before	09.02.2024	
10	Third Mid Term Examinations	06.05.2024	11.05.2024 (1 Week)
11	Summer Vacation	13.05.2024	25.05.2024 (2 Weeks)
12	Submission of Third Mid Term Exam Marks to the University on or before	27.05.2024	
13	Preparation Holidays and Practical Examinations	27.05.2024	01.06.2024 (1 Week)
14	End / Supplementary Examinations	03.06.2024	15.06.2024 (2 Weeks)

Rev
06/7/23
REGISTRAR



Model of evaluated mid exam Answer scripts

College Code : U2

D.V.U.



SREE DATTHA INSTITUTE OF PHARMACY

(Approved by AICTE & PCI, Recg. by Govt. of Telangana., Affiliated to JNTUH)
Sagar Road, Sheriguda, Ibrahimpatnam, R.R. Dist., Greater Hyderabad - 501 510.

ACCREDITED BY NBA



MID EXAM ANSWER BOOK - UG



H.T.No. 2 3 0 2 1 R 0 0 0 1

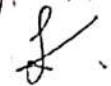
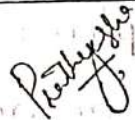
Name of the Exam : mid examination

Name : p.chaitanya

Year / Sem : 1 year / 1 sem Branch : B. Pharmacy Aca. Year : 2023 - 2024

Subject : pharmaceutical inorganic chemistry Date of Exam : 1. 21/12/2023 2.

MARKS AWARDED

MARKS AWARDED										
S U B J E C T I V E	1st M I D	Q.NO.	A	B	C	Total Marks obtained out of 10	O B J E C T I V E	Total Marks obtained out of 10	Invigilator Signature	Evaluator Signature
		1.				10		9		
		2.								
		3.								
	4.									
	2nd M I D	1.				10	9			
		2.								
		3.								
		4.								

Signature of the Principal :

{START WRITING FROM HERE}

Q. Ans

Antacids

Antacids are drugs used to decrease level of gastric acid in the stomach.

- these drugs are used in the case of hyper acidity.
- These drugs are alkaline substances and neutralizing the excess acid in the stomach. the patient suffering from hyperchlorhydria.
- These drugs are give relief pain caused due to hyperchlorhydria.



SREE DATTHA INSTITUTE OF PHARMACY

(Approved by AICTE & PCI, New Delhi, Affiliated to JNTUH, Hyderabad, T.S)
 Nagarjuna Sagar Road, Sheriguda (V), Ibrahimpatnam (M), R.R.Dist., Greater Hyderabad-501510. T.S
 Ph.: +91-8801099936/35, 9393808082,
 Email: principalsdip@sreedattha.ac.in, www.sreedattha.ac.in

Racemic modification :-

A mixture of equal parts of enantiomers is called as Racemic modification. A racemic modification is optically inactive when enantiomers are mixed together. Rotation caused by a molecular of one isomer is exactly counteracted by equal and opposite rotation caused by molecular of its enantiomers. Hence it is used to specific racemic mixture of particular sample.

A racemic modification or racemic is 1:1 mixture enantiomers so that overall optical rotation is zero. The process by which an optically active substance is transformed into the corresponding racemic modification is known as racemisation.

Racemic modification can be prepared in various ways.

* By mixing :- mixing of exactly equal amount of dextrose rotatory (+) and levorotatory (-) isomers. It is a process is generally adopted enantiomers.



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MEDICINAL CHEMISTRY-II

Slip Test - 2

D. Shishu
 2102100067
 3rd Year
 Sec - A

1) Classify ant-neoplastic agents?

Types of Cancer

- * Carcinoma: Originates from epithelial cells
 Ex: Breast Cancer.
- * Sarcoma: Originates from embryonic cells.
- * Blastoma: Originates from embryonic cells.
- * Germ cell cancer: reproductive system related
- * Lymphoma & Leukemia

Classification1) Alkylating agentsa) Nitrogen mustards:

Ex: Mechlorethamine
 Cyclophosphamide
 Melphalon.

b) Ethylenamines:

" " TEPA



→ these drugs are generally weak bases reacts with excess acid in stomach and they convert into salt.

→ these drugs used to increase gastric pH in the stomach

ex:- CaCO_3 , NaHCO_3

→ Neutralization of an antacid is expressed in the "milliequivalent" (mEq) of hydrochloric acid

→ the each antacid have different neutralization capacity. the minimum capacity of antacid is "mEq" of HCl of dosage

classification of Antacids:-

The Antacids are two types:-

1) systemic antacid

2) non-systemic antacid.

1) systemic antacid:-

→ The Antacids are absorbed into systemic circulation (blood circulation) is known as "systemic antacid".

→ These antacids are not suitable and they cause metabolic alkalosis, and effect the Acid-base balance in body.

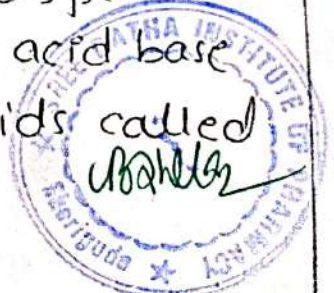
ex:- sodium bicarbonate.

2) non-systemic antacids:-

These antacids are not absorbed into systemic circulation and not able effect the acid base balance in the body. these antacids called

as non-systemic antacids.

ex:- Ammonium chloride.



Ideal properties of Antacids:-

- The Antacids not insoluble in water if they are in particle form
- The Antacids are not cause constipation
- the antacids are not cause any side effect.
- These antacids not cause metabolic alkalosis.
- these are in longer period of time
- These are stable and readily available
- These antacids are not very expensive.

Combination of Antacids:-

The combination of Antacids is the mixing of two antacids one is rapid action and another one is over a longer period of time.

Ex:- combination of magnesium and aluminium
magnesium - rapid action

Aluminium - over a longer period of time.

Ex:-

Sodium bicarbonate:-

molecular formula :- NaHCO_3

molecular weight :- 84.01 g/l

synonym :- Baking soda

Preparation:-

the preparation of sodium bicarbonate is by the solvey ammonia process.



Properties :-

It is a white crystalline powder
It is insoluble in ethanol and soluble in methanol

- It is saline taste
- It is alkaline nature.

Uses :-

- It is used as Antacid.
- It is used as electrolyte replensiner
- It is used as disinfectant.

III II

2Ans)

- ORS means Oral rehydration solution.
- It is also known as oral rehydration therapy.
- It is type of fluid replacement treatment of dehydration caused due to diarrhea.
- It is the more easy and effective treatment of dehydration cause due to diarrhea.
- ORS is have elements, which is lost from our body due to dehydration cause by the diarrhea.
- Glucose, they give through oral route enhance the intestinal absorption of salt and water. Then becomes the not balance the electrolyte.

* Formula of ORS :-

The formula of ORS is given by WHO and UNICEF



2.6 g/L $\rightarrow$ sodium chloride
2.9 g/L $\rightarrow$ trisodium citrate dihydrate
1.5 g/L $\rightarrow$ potassium chloride
13.5 g/L $\rightarrow$ glucose

the total is 20.5 g/L

- $\rightarrow$ Preparation
- $\rightarrow$ The Equipment used in preparation of ORS:-
- $\rightarrow$ Take one litre of boiled and become cooled drinking water
- $\rightarrow$ Take 200ml of clear vessel.
- $\rightarrow$ Take clear vessel to mix the solution
- $\rightarrow$ Take spoon to mix the solution and feed to the child.
- $\rightarrow$ Preparation of ORS
- $\rightarrow$ Take one litre of water
- $\rightarrow$ First wash your hands.
- $\rightarrow$ Take one litre of boiled water.
- $\rightarrow$ (Take) water take into vessel, take ORS packet open the packet and pour the all content into vessel. and mix properly.
- $\rightarrow$ And mix with clear spoon.
- $\rightarrow$ Take spoon and feed to the child in small doses.

2/2



Ans Acid base theories

These theories are designed to explain the concept of acids and bases.

→ These are three types of acid base theories.

1) Arrhenious theory

2) Brounsted Lowry theory

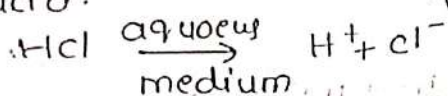
3) Lewis acid-base theory.

1) Arrhenious theory:-

→ It is the mostly common used method for all acids and bases. This theory developed by the scientist Arrhenious. In the year 1884.

→ According to this theory:

→ The substance which release H^+ ions when it dissolved in the aqueous solution, is known as acid.



→ The substance which release OH^- ions when it dissolved in the aqueous medium is known as base

ex:- $NaOH$, $Mg(OH)_2$, $Al(OH)_3$

→ It is a specific scientific theory and simple theory.

Limitations:-

→ This theory will explain the acids and bases present in aqueous medium and fails to explain the in the non-aqueous medium.

→ There are many acids and bases which do not have H^+ and OH^- ions. These are

also called acids and bases

ex:- SO_4^{2-} NH_3 (bases)

BF_3 (acids)

Bronsted Lowry theory:-

- This theory was explained by J.N. Bronsted and J.N. Lowry in 1923
- They gave basic explanation of "Acids and bases."
- According to this theory,
- The acids are which donate proton is known as proton donor. These are called as acids.
- The substance which accepts the proton is known as bases. These are called proton acceptor.

ex:- $\text{HCl} \rightarrow \text{H}^+ \text{Cl}^-$ Acid

$\text{NH}_3 + \text{H}^+ \rightarrow \text{NH}_4^+$ base.

conjugate acid:-

In acid base reactions the reaction b/w acid and base. They form a new base by donating a proton is known as conjugate base.

$\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4^+ \text{Cl}^-$
conjugate base conjugate acid.

conjugate acid

In the p^- which accepts the proton to form conjugate acid.

Limitations:- which do not explain the acids and base which does not contain H^+ and OH^-



ex:- $\text{BF}_3 \cdot \text{SO}_4^{2-}$

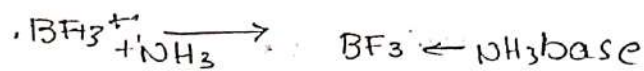
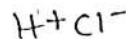
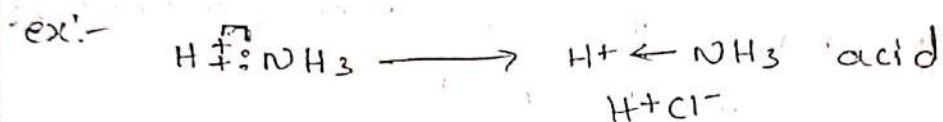
Lewis acid base theory:-

This theory was proposed by Lewis in the year 1930

According to this theory

The acid defines any substance and which accepts electron pair is known to form coordinating bond.

The base defines any substance and which donates electron pair to form coordinate bond.



Limitations:-

This theory which fails to explain the substance does not involve in formation of coordinate covalent bond.

✓
✓



MTD Exam - II

Ans

Antimicrobials.

Antimicrobials are drugs which are used to destroy or kill the (microorganisms) growth of microorganisms.

→ It is also kill the growth of bacteria.

→ Antimicrobial contains

- 1) Antiseptic
- 2) Disinfectant
- 3) Germicides
- 4) Keratolytics
- 5) Sanitizers
- 6) Sterilization

1) Antiseptic :-

Agents which are used to kill or inhibit the growth of microorganisms, which are applied topically to kill microorganisms.

ex:- Sodium chloride, Ammonium chloride.

2) Disinfectant :-

Agents which are used to kill or inhibit the growth of microbes on the objects, equipments, etc.

ex:- Sodium chloride.

3) Germicides.

Agents which inhibit the growth of microbes in living and non-living substances.

Bactericides → inhibit growth of Bacteria

Fungicides → inhibit growth of fungi

Virucides → inhibit growth of virus.



4. Keratolytics:-

these are agents which ^{inhibit} growth of microorganisms and these applied on skin & soothing in nature.

5. skin sanitizers.

→ these are agents which inhibit growth of microbes these are used in cleaning and washing purpose.

6. sterilization:-

It is a complete killing of microbes. By these process the microbes will be killed and inhibited.

→ microorganisms contain protein. Based on proteins the microorganisms can survive.

→ Antimicrobials change their protein structure and kill the microorganisms.

Mechanisms of actions:

1) Oxidation

2) Halogenation.

3) Precipitation.

Oxidation.

→ The mechanism of action contains oxygen molecules. By the oxidation, these molecules focus on functional groups and kill the growth of microorganisms.

Halogenation

→ The mechanism of action contains halogen compounds. The halide compounds attach to the focus on main function and kill the growth of microorganisms.

Precipitation

The mechanism of action contains



→ The metal ions are focused on main functional group and inhibit the growth of microorganisms.
Ex:- Hydrogen peroxide

Hydrogen peroxide

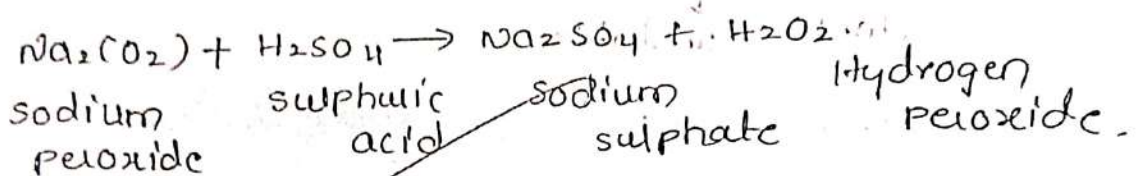
Chemical formula :- H_2O_2

molecular weight :- 34.01 gm

Synonym :- Hydrogen oxide.

Preparation of hydrogen peroxide.

It can be prepared by sodium peroxide with sulphuric acid



Assay of hydrogen peroxide.

Take the accurate amount of 10gm of sodium peroxide and dissolve in some of water and add sulphuric acid 0.5gm. and titrate and potassium permanganate as an indicator.

Properties :-

It is slightly acidic nature.

It is saline taste.

It is odourless.

It is soluble in water, insoluble in alcohol.

Uses :-

→ It is used as antiseptic

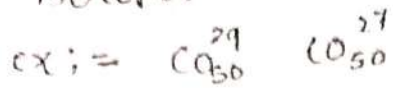
→ It is used as antimicrobial.

→ It is used in cutting, wounds cleaning.

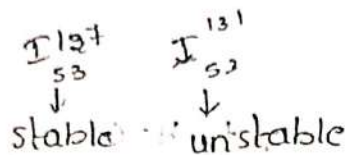
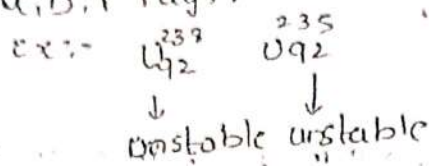
→ It is also used in disinfectant.



20) Isotopes :- the isotopes contains same atomic number but different atomic mass is known as isotopes



Radioisotopes :- these are Radioisotopes contains unstable nucleus and emits electrons in the form of α, β, γ rays.



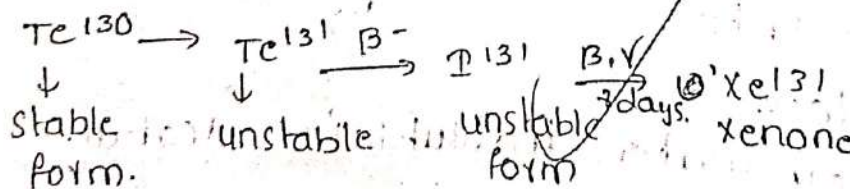
Ex :- sodium iodide

sodium iodide

Chemical formula :- I_{131}

Synonyms :- Radio active chemical

Preparation :- The preparation of I_{131} is prepared in a nuclear reaction of neutron irradiation of natural tellurium to get target almost I_{131} is Radio halide. It is longer than half life.



Properties :-

It's pH is ≈ 7 (3-4), (7-9)

$t_{1/2} = 8 \text{ days}$

It emits β and γ rays



It is clear colourless.

uses:-

- It is used in treatment of graves disease
- It is detection of thyroid diseases
- It is used in treatment of thyrotoxicosis
- It is used in treatment of thyroid disease.

1Ans. Expectorant

Expectorants are drugs (used in) which increase fluidity of sputum and volume of sputum, the sputum is removed out by coughing.

The Expectorant are two types.

1) sedative expectorants

2) stimulant expectorants

Sedative expectorants.

These which cause stomach irritant and in the stomach, cause the gastric reflex. These drugs are increasing the flu, volume of sputum. ^{these drugs} increase ^{sedation} ex: = Indian quill, ammonium chloride, and sodium citrate.

stimulant expectorants:-

These expectorants stimulate fluid in the respiratory tract and are directly or indirectly by increasing the fluidity of sputum, easily discharge from respiratory tract.

ex: = lemon, Asparagus.



Ex:- Ammonium chloride :-

Ammonium chloride :-

Ammonium chloride (NH_4Cl)

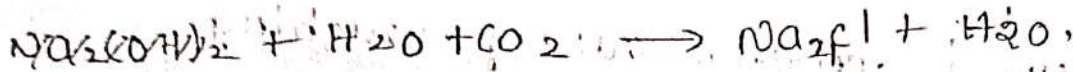
molecular weight :- 53.49 gm

Synonym :- sal ammoniac, ammoniamuriate.

It is highly in volcano places.

Preparation :-

→ It can be prepared by sal ammoniac solvey ammonia process.



→ $2(\text{NH}_3)$ ammonia chloride

It also prepared by double

Assay :-

Take accurately weighed ammonia chloride and add N/10 silver nitrate and add sodium thiocyanate and ferric sulphate is an indicator and

Properties

It is odourless

It is tasteless

It is white colour powder

It is used for saline taste

uses :-

→ It is used as expectorant

→ It is used as Acidifier.

→ It is used as cathetics



2-10-15

Half life:-

Half life is defined as time ^{required to} of Radio active chemical it decay into the one half of its original value is called Half life

$t_{1/2} \rightarrow$ It is denoted in $t_{1/2}$

$$\frac{0.693}{\lambda} \rightarrow \text{Integration constant.}$$

Half life of Elements

Polonium 131 $\rightarrow 3 \times 10^7 \text{ sec}$

$\text{I}_{131} \rightarrow 7 \text{ days}$

$^{227}\text{Ac} \rightarrow 60 - 150 \text{ day}$

$\text{P}_{23} \rightarrow 143 \text{ days}$

$\text{N}_{22} \rightarrow 2.6 \text{ years}$

$\text{U}_{238} \rightarrow 4.5 \times 10^9 \text{ years}$

22



Model of sample slip test papers



MEDICINAL CHEMISTRY-II

Slip Test - 2

D. Shisouha
21021R0067

3rd Year
Sec - A

1) Classify ant-neoplastic agents?

1. Types of Cancer

* Carcinoma: Originates from epithelial cells

Ex: Breast Cancer.

* Sarcoma: Originates from embryonic cells.

* Blastoma: Originates from embryonic cells.

* Germ cell Cancer: reproductive system related.

* Lymphoma & Leukemia

Classification

1) Alkylating agents

a) Nitrogen mustards:

Eg: Mechlorethamine
Cyclophosphamide
Melfhalan.

b) Ethylenamines:

Eg: thio-TEPA



c) Alkyl Sulphonate:

Eg: Busulfan

d) Nitrosourea:

Eg: Carmustine, lomustine

e) Triazine:

Eg: Dacarbazine

2) Antimetabolites

a) Folate antagonist

Eg: Methotrexate

b) Purine antagonist

Eg: 6-mercaptopurine

6-thioguanine

Azathioprine

c) Pyrimidine antagonist

Eg: 5-fluorouracil

Fluorouridine

Cytarabine

d) Vinca alkaloids

Eg: Vincristine

3) Taxane:

Eg: Paclitaxel

4) Epidolophyllotoxins

Eg: Etoposide

5) Antibiotics

Eg: Actinomycin

Doxorubicin

Daunorubicin

Bleomycin

6) Miscellaneous:

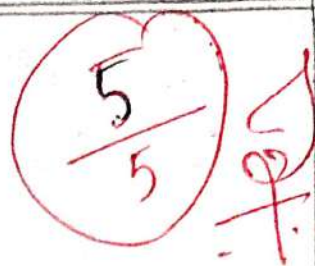
Eg: Cisplatin

Mitomycin

Model of Assignment papers

CS PHA

Assignment



subject:- pharmaceutical organic chemistry - III

Topics:- Racemic modification
Resolution of Racemic mixture

Reference:- Arun Bahl & Bahl

submitted to:- Prathyusha mam

Name:- C. Ajay

Roll no:- 22U21R0036

section:- A

Batch:- II



Racemic modification :-

A mixture of equal parts of enantiomers is called as Racemic modification. A racemic modification is optically inactive when enantiomers are mixed together. Rotation caused by a molecule of one isomer is exactly counteracted by equal and opposite rotation caused by molecule of its enantiomer. The prefix it is used to specific racemic mixture of particular sample.

A racemic modification or racemic is 1:1 mixture enantiomers so that overall optical rotation is zero. The process by which an optically active substance is transformed into the corresponding racemic modification is known as racemisation.

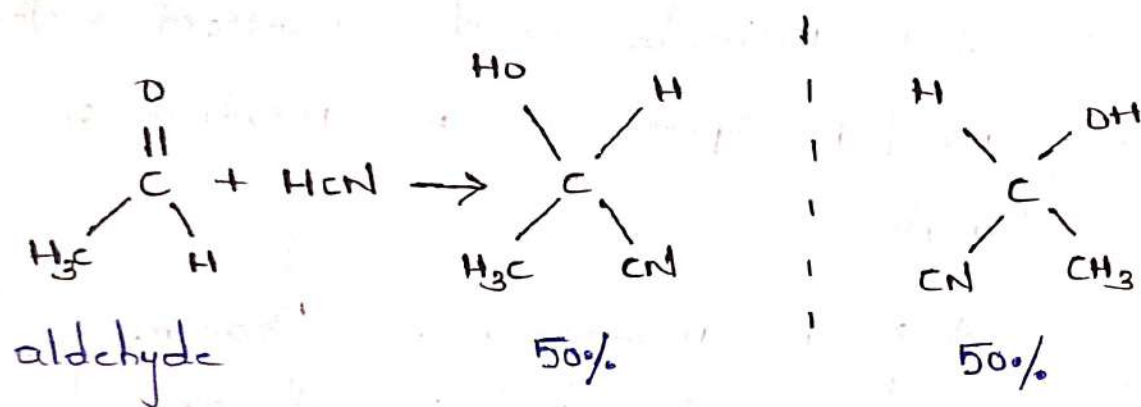
Racemic modification can be prepared in various ways.

* By mixing :- mixing of equal amount dextrose rotatory (+) and levorotatory (-) isomeric. It is a process is generally adopted enantiomers



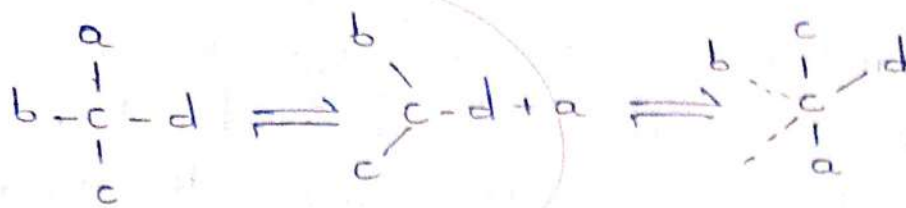
is both the cases in belong the same species
will have same mp.

* By Synthesis :- Synthesis of asymmetric compound from symmetrical compound in absence of optical active produce a racemic.



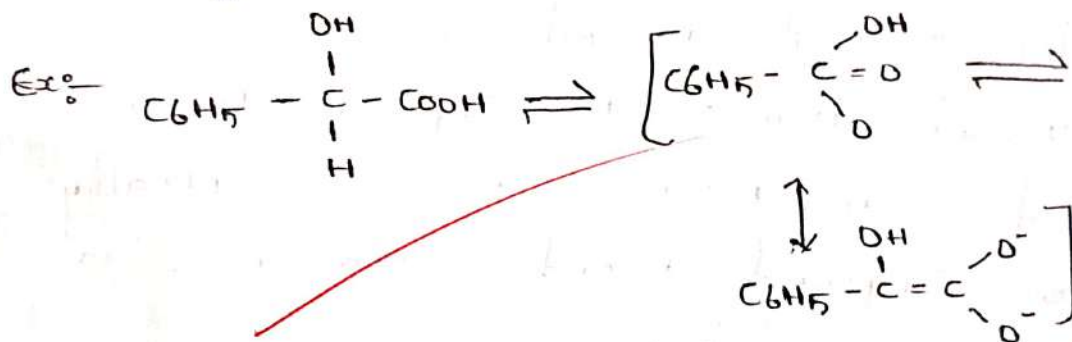
* By racemization :-

Heating at higher temperature is homolytic splitting of one of the 4 bond attached to chiral 'c' atom separated a group exchanges place with one of the remaining groups enantiomers of original is to be produced.

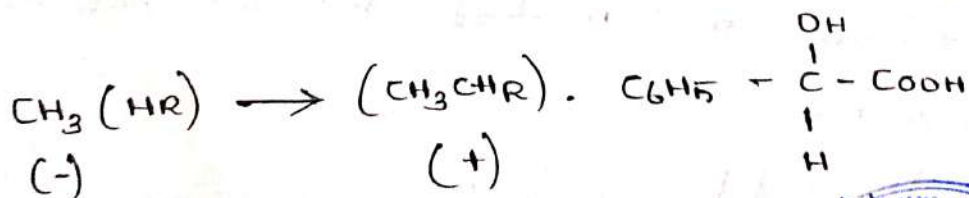


b) By un-ion formation :-

It involves heterolytic cleavage, so that the bonds breaks with R R' R' group. Thus, is converted to an anion H⁺ is separated c. By cation formation it involves temporarily removal of group with pow of e⁻ to a level racemisation is brought.

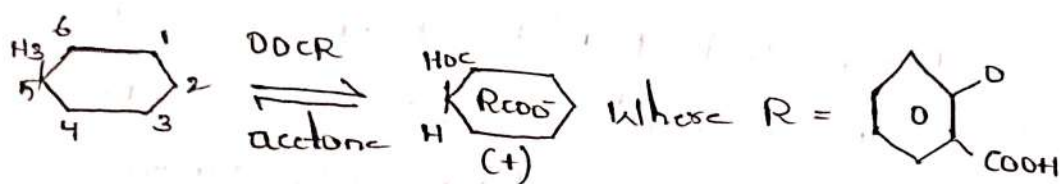


By Warden inversion. Inversion of Configuration.



By chemical transformation :- It is possible to change a dissymmetric molecule to its enantiomers without ever breaking any of the bonds to the asymmetric atom. Conversion of enantiomers to the bonds

5-methyl-2-cyclohexenylphtale



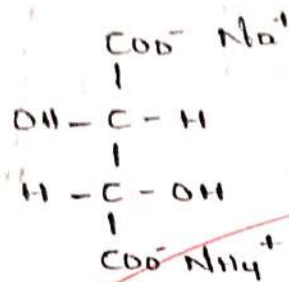
→ Resolution of Racemic mixture :-

Synthesis of an optically active compound a mixture of both (+) and (-) isomers in equal amounts they are non-superimposable mixture is called racemic mixture into its 2 optically active enantiomers (+) and (-) is known as a resolution.

Following are 3 methods for resolving an enantiomeric pair.

1. Mechanical Resolution :- Pasteur's original separation of a NaNH_4^+ salt of tartaric acid known as mechanical resolution.

By a slow crystallization of this salt. He obtained a crystalline form resulted in the resolution of a racemic mixture not generally practical because most racemic mixture do not readily separate and the separation would be extremely tedious.



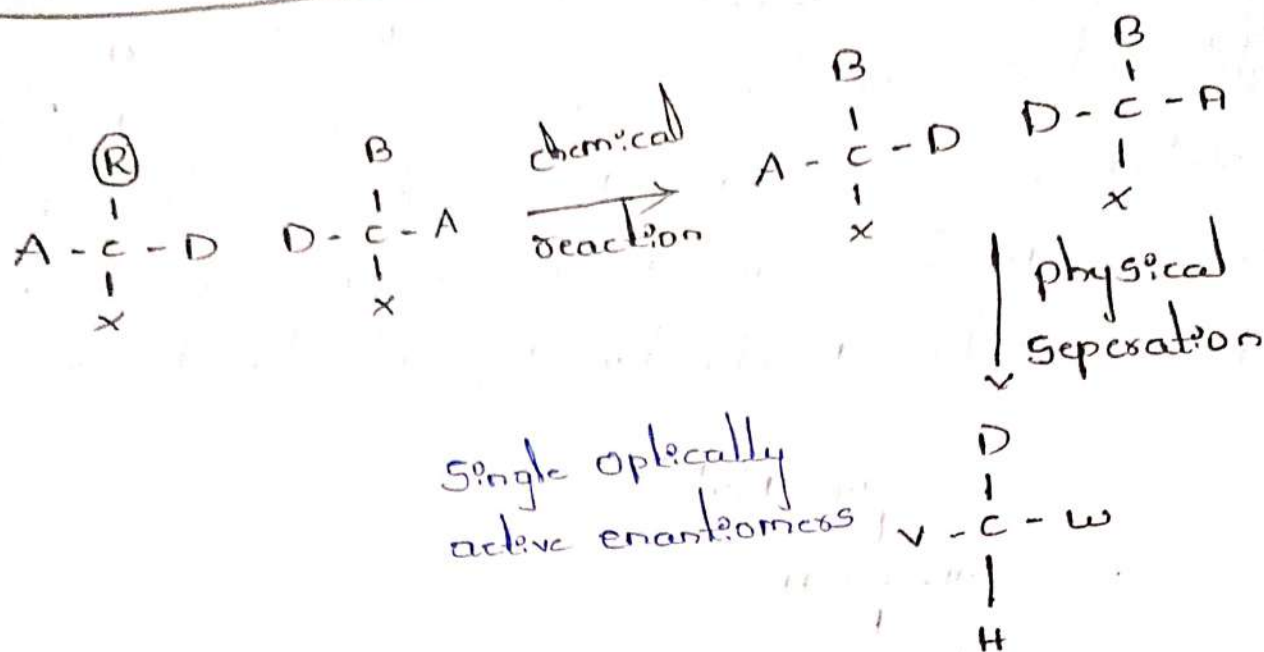
2. Biological resolution :-

This method takes advantages of the varying rates of metabolism of compounds of a racemic mixture by a micro-organisms produce enzymes that are themselves chiral and consequently differentiate with a enantiomeric forms.

3. Resolution through Diastereomers :-

The most useful method for separating enantiomers involves converting them into a pair of diastereomers. A racemic mixture can be converted into a pair of diastereomers.





4. Complex Formation :-

Sometimes racemates form crystallization complex with a diagram reagent of diastereomeric complex passes a solubilities on will ppt. After the 2 original enantiomers can be obtained from a 2 complexes

* Asymmetric Synthesis :-

In this method, optically active compound is directly prepared from an optically inactive compound under the influence of a some other optically active substance.

Racemic mixture + Optically active $\rightarrow$ Optically active

$\left\{ \begin{array}{l} 50\% \text{ Dextro} \\ 50\% \text{ Levo} \end{array} \right\}$

Optically inactive

$\left\{ \begin{array}{l} \text{Dextro/Levo} \\ \text{anyone} \end{array} \right\}$

Optically active

* It is two types :-

- i) partial asymmetric synthesis
- ii) Absolute asymmetric synthesis

Optically inactive Compound $\xrightarrow{\text{Asymmetric Synthesis}}$ Optically active Compound.

* partial asymmetric symmetric :-

In this both (+) & (-) both are formed.

Eg :- glucose etc

50% 50% $\xrightarrow{\text{Optically active}}$ 10% (-)

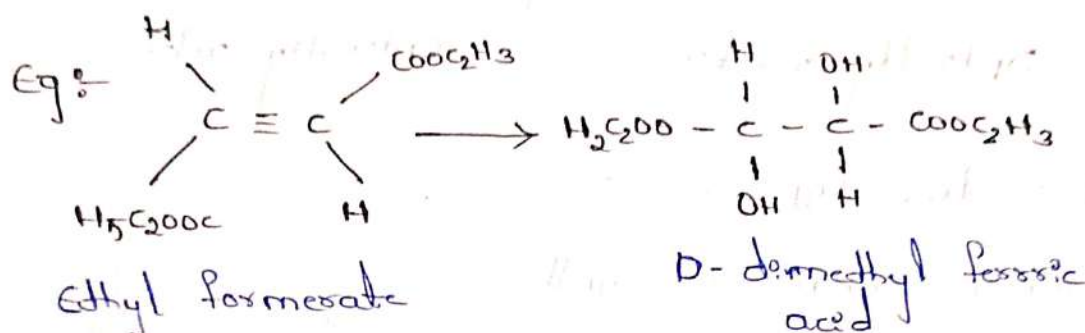
Racemic mixture

Optically active



ii) Absolute asymmetric synthesis :-

In this synthesis, racemic mixture is completely convert into optically active compound any one



* Ethyl fumarate and H_2O_2 under a influence of centros rotatory light give (+) diethyl tartaric acid.



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD
HYDERABAD-500085

SREE DATTHA INSTITUTE OF PHARMACY(U2)
B.Pharmacy - R22 - I Year - I Semester
B.Pharmacy

Final University Consolidated Internal Marks Report-Date- 2024-02-29 16.59.42

HNTU	28101	28102	28103	28104	28105	281AA	281AB	281AC	281AD	281AE	281AG
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23U21R0002	11	14	15	11	12	21	21	22	21	23	22
23U21R0003	11	15	15	14	14	21	21	21	21	21	20
23U21R0004	13	15	15	14	14	23	24	23	23	24	21
23U21R0005	12	15	15	14	14	21	24	24	24	24	23
23U21R0006	12	14	15	13	13	22	23	24	24	24	22
23U21R0007	12	15	15	13	14	24	25	24	23	24	21
23U21R0008	11	14	15	12	12	22	21	21	20	21	18
23U21R0009	13	15	14	14	13	23	22	23	22	23	20
23U21R0010	13	15	14	11	13	22	23	23	22	22	21
23U21R0011	11	12	13	10	12	18	18	19	19	20	18
23U21R0012	15	13	15	14	12	23	24	25	23	24	21
23U21R0013	13	15	15	14	14	25	25	24	23	24	22
23U21R0014	14	14	15	13	14	25	25	23	24	24	23
23U21R0015	14	15	15	13	13	24	25	24	23	24	22
23U21R0016	13	15	15	14	14	23	25	24	23	23	21
23U21R0017	14	15	15	14	13	23	24	24	23	24	23
23U21R0018	12	15	14	13	14	23	23	24	22	24	21
23U21R0019	12	15	15	12	13	20	22	24	19	23	21
23U21R0020	13	15	15	12	12	20	18	23	18	23	19
23U21R0021	13	15	15	13	13	24	24	23	23	24	21
23U21R0022	14	15	15	14	14	24	25	25	24	24	23
23U21R0023	13	15	14	12	12	22	23	24	22	22	22
23U21R0024	11	15	15	14	13	24	25	24	24	22	22
23U21R0025	10	15	15	14	13	20	25	24	22	23	20
23U21R0026	13	15	15	14	14	24	25	24	21	23	23
23U21R0027	14	15	15	12	14	23	25	23	22	24	23
23U21R0028	13	15	15	12	13	23	25	23	21	22	22
23U21R0029	12	13	14	11	12	20	21	21	19	22	20
23U21R0030	14	14	15	12	13	21	23	20	21	22	20
23U21R0031	13	14	15	12	14	22	25	23	21	22	20
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HNT0	28101	28102	28103	28104	28105	281AA	281AB	281AC	281AD	281AE	281AG
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23U21R0034	13	14	15	13	13	22	25	23	21	22	21
23U21R0035	13	15	15	12	14	20	25	23	22	22	23
23U21R0036	14	14	15	12	13	22	25	22	22	23	22
23U21R0037	13	14	13	10	12	19	21	20	19	21	19
23U21R0038	12	15	15	12	12	21	25	23	21	23	21
23U21R0039	12	15	13	10	12	18	20	20	18	20	21
23U21R0040	13	15	14	11	14	21	24	22	23	22	19
23U21R0041	13	15	13	12	12	18	19	20	18	20	19
23U21R0042	13	15	15	12	14	21	24	22	22	22	21
23U21R0043	11	15	12	10	13	18	18	20	19	18	19
23U21R0044	12	15	14	12	14	21	25	22	21	23	21
23U21R0045	13	15	15	12	13	22	24	21	19	21	21
23U21R0046	12	15	13	10	12	19	21	19	19	21	19
23U21R0047	13	15	14	12	12	21	22	21	21	22	20
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23U21R0051	13	15	15	12	14	20	23	20	21	22	19
23U21R0052	14	15	15	12	12	19	23	20	23	22	21
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23U21R0054	11	13	13	10	12	18	19	22	20	20	20
23U21R0055	10	12	11	12	12	20	18	21	19	19	18
23U21R0056	13	15	15	12	14	23	23	21	21	23	20
23U21R0057	12	15	15	12	13	22	21	21	22	23	23
23U21R0058	11	15	15	10	13	18	21	20	21	20	21
23U21R0059	14	15	15	12	14	20	24	22	23	23	22
23U21R0060	14	15	15	12	14	19	25	23	24	24	22
23U21R0061	13	14	15	10	14	20	23	23	22	23	21
23U21R0062	11	13	14	12	12	18	22	23	23	21	22
23U21R0063	12	14	15	12	14	24	25	23	24	23	23
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23U21R0066	13	12	15	13	14	25	25	23	23	23	21
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23U21R0068	12	15	15	12	14	19	22	22	21	22	21
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23U21R0070	10	14	14	12	12	18	20	21	23	20	19
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23U21R0072	11	15	15	13	14	22	24	22	23	22	23
23U21R0073	14	15	15	12	14	22	23	20	22	23	20
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23U21R0075	13	13	14	10	13	21	19	19	18	22	21
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23U21R0077	12	13	15	10	13	23	19	19	18	20	22



HNT0	28101	28102	28103	28104	28105	281AA	281AB	281AC	281AD	281AE	281AG
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23U21R0080	13	15	15	13	13	23	24	22	22	22	20
23U21R0081	13	15	15	14	14	21	24	24	23	23	21
23U21R0082	14	15	15	13	13	21	23	21	21	21	22
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23U21R0089	12	13	15	12	13	21	20	19	22	22	21
23U21R0090	12	15	15	12	14	24	21	21	21	22	21
23U21R0091	13	14	15	12	13	21	20	19	21	23	21
23U21R0092	10	14	15	10	12	22	23	18	19	21	20
23U21R0093	13	14	15	13	13	21	20	20	20	21	22
23U21R0094	10	11	11	10	12	15	15	15	15	15	15
23U21R0095	10	11	11	10	12	15	15	15	15	15	15
23U21R0096	11	14	14	10	12	20	19	18	18	19	21
23U21R0097	12	13	15	13	14	22	18	21	21	21	20
23U21R0098	11	12	13	10	12	21	18	21	21	19	20
23U21R0099	11	13	13	13	12	20	20	23	22	21	22
23U21R00A0	10	12	13	10	12	21	18	22	21	20	20
23U21R00A1	12	13	15	12	13	21	18	23	20	19	20
23U21R00A2	12	12	13	10	12	20	18	20	21	19	20
23U21R00A3	12	13	15	13	13	20	20	19	21	19	20
23U21R00A6	10	11	11	10	12	15	15	15	15	15	15
Total:104	128 6	147 7	150 0	125 7	135 8	221 2	229 9	223 7	220 7	226 2	216 3

Note : '-1' indicates student is absent for the exam.

Subject Code	Subject Name
281AA	COMMUNICATION SKILLS
28103	PHARMACEUTICAL ANALYSIS I LAB
281AD	PHARMACEUTICAL INORGANIC CHEMISTRY
28104	PHARMACEUTICAL INORGANIC CHEMISTRY LAB
281AE	PHARMACEUTICS
28101	COMMUNICATION SKILLS LAB
281AB	HUMAN ANATOMY AND PHYSIOLOGY
281AC	PHARMACEUTICAL ANALYSIS I
28102	HUMAN ANATOMY AND PHYSIOLOGY I LAB



28105	PHARMACEUTICS LAB
281AG	REMEDIAL MATHEMATICS

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Jawahar Lal Nehru Technological University Hyderabad
Kukatpally, Hyderabad - 500 085, Telangana, India
Online Registrations

SREE DATTA INSTITUTE OF PHARMACY(U2)

Forfeight Attendance	Even Registrations	Overns, Seating and Nominal Rolls	RC Fee Registration	Upload Faculty Details	Midwise Internal Entry	Lab External Marks Entry	Reports	Admin options
					Midwise Internal Marks Entry/Edit for Theory			
					Lab/Seminar/Mandatory Internal Marks Entry			
					Generate Midwise PDF Report			
					Generate Pdf Consolidate Report			
					Supply Manualatory Subjects/Seminar InternalMarks Entry			
					Generate PDF for Supply Mandatory/Seminar Subjects Marks			
					Upload NSS/NSO Participation Details			
					Generate NSS/NSO Participation Details PDF Report			





JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD
HYDERABAD-500085

SREE DATTHA INSTITUTE OF PHARMACY(U2)
M.Pharmacy - R22 - I Year - I Semester
PHARMACEUTICAL REGULATORY AFFAIRS

Final University Consolidated Internal Marks Report-Date- 18-03-2024 18.33.16

HTNO	681301	681302	6813AA	6813AB	6813AE	6813AH	6813AJ	6813AN	A681303	S681303
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23U21S1302	22	21	21	20	21	21	21	86	46	47
23U21S1303	21	21	21	21	21	21	20	85	45	46
23U21S1304	22	22	21	21	21	21	21	85	45	45
23U21S1305	21	21	21	20	21	21	21	86	46	46
23U21S1306	22	21	20	21	21	21	20	85	45	45
23U21S1307	21	22	21	20	21	21	21	85	45	46
23U21S1308	22	21	21	21	20	21	20	86	46	46
23U21S1309	21	22	20	20	21	21	21	85	45	45
23U21S1310	22	21	21	21	21	21	21	86	45	46
23U21S1311	21	21	21	21	21	21	21	85	45	46



HTNO	681301	681302	6813AA	6813AB	6813AE	6813AH	6813AJ	6813AN	A681303	S681303
23U21S1312	21	22	21	21	21	21	21	86	45	46
23U21S1313	21	21	21	21	21	21	21	85	45	46
23U21S1314	22	21	21	21	21	21	21	85	45	46
23U21S1315	21	22	20	21	21	21	20	85	45	45
Total:15	321	321	311	311	314	315	311	128	678	687

Subject Code	Subject Name
6813AN	ENGLISH FOR RESEARCH PAPER WRITING
681301	DRUG REGULATION AND REGISTRATION LAB
6813AB	GOOD REGULATORY PRACTICES
6813AA	DRUG REGULATORY AFFAIRS
681302	REGULATORY PRACTICE AND DOCUMENTATION LAB
6813AE	TOTAL QUALITY MANAGEMENT
6813AH	STABILITY OF DRUGS AND DOSAGE FORMS
6813AJ	RESEARCH METHODOLOGY AND IPR
S681303	SEMINAR MARKS
A681303	ASSIGNMENT MARKS



Signature Of Principal with date & Office seal



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD
HYDERABAD-500085

SREE DATTHA INSTITUTE OF PHARMACY(U2)
M.Pharmacy - R22 - I Year - I Semester
PHARMACEUTICS

Final University Consolidated Internal Marks Report-Date- 18-03-2024 18.30.24

HTNO	680301	680302	6803AA	6803AB	6803AE	6803AH	6803AJ	6803AN	A680303	S680303
23U21S0301	23	23	21	21	21	21	21	88	46	46
23U21S0302	23	24	21	21	20	21	21	87	46	45
23U21S0303	24	24	21	21	21	20	21	87	46	46
23U21S0304	23	24	21	21	21	21	21	86	46	46
23U21S0305	24	23	21	20	21	21	21	86	45	45
23U21S0306	23	24	21	21	21	21	21	85	46	46
23U21S0307	24	23	21	21	20	21	21	87	46	46
23U21S0308	23	23	21	21	21	20	20	85	46	45
23U21S0309	24	24	20	21	21	21	21	86	46	46
23U21S0310	23	23	21	20	21	21	21	85	45	46
23U21S0311	24	24	21	21	21	21	21	86	46	45



HTNO	680301	680302	6803AA	6803AB	6803AE	6803AH	6803AJ	6803AN	A680303	S680303
23U21S0312	23	23	21	21	21	21	21	85	46	46
23U21S0313	24	24	21	21	20	21	21	84	45	45
23U21S0314	23	23	21	21	21	21	21	85	46	46
23U21S0315	24	24	21	21	21	21	21	85	46	45
23U21S0316	23	23	21	21	20	20	20	86	45	46
Total 16	375	376	335	334	332	333	334	137 3	732	730

Subject Code	Subject Name
6803AN	ENGLISH FOR RESEARCH PAPER WRITING
6803AA	APPLIED BIOPHARMACEUTICS AND PHARMACOKINETICS
680302	MODERN PHARMACEUTICS I LAB
680301	APPLIED BIOPHARMACEUTICS AND PHARMACOKINETICS LAB
6803AB	MODERN PHARMACEUTICS I
6803AE	TOTAL QUALITY MANAGEMENT
6803AH	STABILITY OF DRUGS AND DOSAGE FORMS
S680303	SEMINAR MARKS
6803AJ	RESEARCH METHODOLOGY AND IPR
A680303	ASSIGNMENT MARKS



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JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD
HYDERABAD-500085

SREEDATHA INSTITUTE OF PHARMACY(U2)
Pharm.D - R08 - I Year
PHARM D

Final University Consolidated Internal Marks Report-Date- 2024-07-30 13.49.41

HTNO	PH101	PH102	PH103	PH104	PH105	PH106	PH107	PH108	PH109	PH110	PH111	PH112	PH113
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23U21T0002	25	30	22	30	28	30	-	29	29	27	30	30	-
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23U21T0004	28	30	26	30	30	30	-	29	30	30	30	29	-
23U21T0005	20	21	20	20	20	21	-	26	24	26	27	29	-
23U21T0006	22	24	23	29	27	28	-	28	28	28	30	29	-
23U21T0007	24	30	27	30	25	28	-	28	29	24	30	29	-
23U21T0008	28	29	27	30	28	30	-	28	27	25	30	29	-
23U21T0009	26	24	23	30	27	28	-	28	28	30	29	29	-
23U21T0010	20	24	23	30	29	29	-	28	30	26	30	29	-
23U21T0011	25	25	22	29	28	28	-	26	24	24	2	28	-
23U21T0012	27	28	24	30	28	29	-	29	27	26	30	29	-
23U21T0013	26	26	26	30	28	29	-	29	30	25	29	29	-
23U21T0014	23	20	20	21	20	20	-	27	25	23	27	27	-



HTNO	PH101	PH102	PH103	PH104	PH105	PH106	PH107	PH108	PH109	PH110	PH111	PH112	PH113
23U21T0015	26	30	26	30	29	30	-	29	30	30	30	30	-
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23U21T0017	27	28	28	30	29	30	-	29	27	26	30	29	-
23U21T0018	23	21	20	27	23	23	-	25	26	24	27	26	-
23U21T0019	27	26	22	29	26	28	-	28	28	26	29	29	-
23U21T0020	18	20	20	26	22	30	-	24	25	20	27	25	-
23U21T0021	24	27	25	30	29	28	-	28	29	26	29	29	-
23U21T0022	20	23	20	28	21	26	-	28	28	24	29	28	-
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23U21T0024	25	29	26	30	30	30	-	29	26	25	30	28	-
23U21T0025	23	27	25	30	28	27	-	29	27	25	29	30	-
23U21T0026	24	22	20	25	21	21	-	27	23	23	27	26	-
23U21T0027	22	24	21	27	23	25	-	29	27	22	20	26	-
23U21T0028	24	29	23	30	28	29	-	29	30	24	30	30	-
23U21T0029	22	21	20	27	20	26	-	28	23	21	30	27	-
23U21T0030	24	26	23	27	26	30	-	29	28	25	30	29	-
23U21T0031	22	25	20	27	23	27	-	26	25	25	30	29	-
23U21T0032	26	21	20	28	20	30	-	28	25	25	29	29	-
Total:32	77	81	74	90	82	88	0	89	87	80	89	91	0
	2	3	1	3	5	6		6	1	8	8	0	

** indicates that Marks exceeding the Max.marks

Subject Code	Subject Name
PH107	REMEDIAL BIOLOGY
PH108	HUMAN ANATOMY & PHYSIOLOGY LAB
PH109	PHARMACEUTICS LAB
PH110	MEDICINAL BIOCHEMISTRY LAB



PH111	PHARMACEUTICAL ORGANIC CHEMISTRY LAB
PH101	HUMAN ANATOMY & PHYSIOLOGY
PH112	PHARMACEUTICAL INORGANIC CHEMISTRY LAB
PH102	PHARMACEUTICS
PH113	REMEDIAL BIOLOGY LAB
PH103	MEDICINAL BIOCHEMISTRY
PH104	PHARMACEUTICAL ORGANIC CHEMISTRY
PH105	PHARMACEUTICAL INORGANIC CHEMISTRY
PH106	REMEDIAL MATHEMATICS

Signature Of Principal with Office seal



SREE DATTHA INSTITUTE OF PHARMACY(U2)-B.Pharmacy List Of Students Applied For RCRV

* RC/RV- RC Stands For Recounting / RV Stands for Revaluation

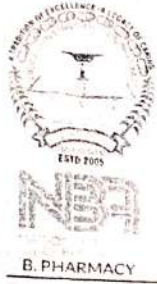
S.NO	HTNO	Regulation	Year	Semester	SubjectName(SubjectCode)	Applied For	Amount
1	20U21R0072	R17	4	2	SOCIAL AND PREVENTIVE PHARMACY(248AB)	RV	1000
2	20U21R0084	R17	4	2	NANO TECHNOLOGY(248AE)	RV	1000
3	20U21R0076	R17	4	2	NANO TECHNOLOGY(248AE)	RV	1000
4	20U21R0076	R17	4	2	SOCIAL AND PREVENTIVE PHARMACY(248AB)	RV	1000
5	20U21R0084	R17	4	2	SOCIAL AND PREVENTIVE PHARMACY(248AB)	RV	1000

Total RV Amount:5000

Total RC Amount:0

Grand Amount:5000





JNTUH COLLEGE CODE : U2

EAMCET CODE : SDIP

SREE DATTHA INSTITUTE OF PHARMACY

(Accredited by NBA & NAAC, Approved by AICTE & PCI, New Delhi, Affiliated to JNTUH, Hyderabad, T.S.)
Nagarjuna Sagar Road, Sheriguda(V), Ibrahimpatnam (M), R.R. Dist. Greater Hyderabad - 501510. T.S.

Ph. : +91 - 8801099936 / 35, 9393808082

Email : principalsdip@sreedattha.ac.in, www.sreedattha.ac.in



Ref. No.

062

Date : 27-07-2023

31 JUL 2023

Dr. S.A. Sreenivas,
M.Pharm, Ph.D.
Professor & Principal.

To
The Controller of Examination,
JNTU-H,
Kukatpally,
Hyderabad.



Dear Sir/Madam,

Sub: Mr. RAVULA GOPI bearing JNTUH No. 19U21R0012- Grace
Marks....Reg .

With reference to above subjected, we bring to your kind information that **Mr. RAVULA GOPI** bearing JNTUH No. 19U21R0012 is a bonafide student studying B.Pharmacy in our college. He has passed in all the subjects except Medicinal Chemistry of B.Pharm. II Year II semester. He is applying for the **Grace Marks** based on the result received in March 2022 examination. We are enclosing the marks memo for your kind perusal.

Kindly do the needful and oblige.

Thanking You,


Dr. S A SREENIVAS
M. Pharm, Ph. D

PRINCIPAL

Sree Dattha Institute of Pharmacy
Sheriguda V Ibrahimpatnam, R.R. Dist., 501 510.

Encl: II/II Marks Memo



Sponsored By : Vyjayanthi Educational Society

FORM-5

Date: 24/07/2023

I, RAVULA GOPI bearing the HT No 19U21R0012 student of
Sree Datta Institute of Pharmacy college Studied B.Tech./B.Pharm. during the
A.Y. 2019-2023 failed in subjects (limited to four subjects) and addition of a few marks will enable
me to pass in the failed subject(s). The details of the failed subject(s) for which award of grace marks is
requested is given below.

S.No	Year & Semester	Subject code	Subject Name	Internal Marks	External marks secured in recent exam
1	II nd yr - II Sem	244AB	Medicinal Chemi-stry-1	20	19

I hear by submit that I did not apply for supplementary examination and will not apply in future for any supplementary examination. Hence, I request for processing of my Grace Marks application. Further I request you to issue me my Consolidated Marks Memo and Provisional Certificate, at the earliest, after adding grace marks.


Signature of the student

Enclosure: Original Marks Memo(s) containing subject(s) for which award of Grace Marks requested.

INSTRUCTIONS: The application along with the necessary documents is to be submitted at the counter of Student Service Section, JNTUH Examination Branch

To
The Director of Evaluation
JNTUH
Kukatpally, Hyderabad - 500085

Sir,

Sub: Request for processing of award of Grace Marks and issue of PC & CMM - Reg. Ref:
Hall-ticket number of the candidate is 19U21R0012

With reference to the subject cited above, I here by forward the application for award of Grace marks of Mr./Ms. RAVULA GOPI, a student of B.Tech. bearing the hall-ticket number 19U21R0012.




Signature of Principal



Dr. S A SREENIVAS
M. Pharm, Ph. D
PRINCIPAL
Sree Datta Institute of Pharmacy
Sherguda, Nizamapattanam, R.R. Dist., -501 510



JAWAHARLAL NEHRU TECHNOLOGICAL UNIVERSITY HYDERABAD
KUKATPALLY, HYDERABAD - 500 085, TELANGANA STATE, INDIA.



MEMORANDUM OF GRADES

Memo. No.: IM 5624374

Serial No.: 10774004493

Examination: B.Pharmacy II Year II Sem. (R17) Suppleme

Branch:

Name: RAVULA GOPI



Roll Ticket No.:

19U21R0012

Month & Year of Exam: March, 2022

Name of the College: U2-S D I P, IBRAHIMPATNAM

S.No.	SUBJECT CODE	SUBJECT TITLE	Grade Secured	Grade Point(Gi)	Result	Credits (Ci)
1	244AB	MEDICINAL CHEMISTRY - I	F	0	F	0.0
2	244AE	PHARMACOGNOSY AND PHYTOCHEMISTRY - I	B	6	P	4.0
SUBJECTS REGISTERED: 2 APPEARED: 2 PASSED: 1 TOTAL:						04.0
Semester Grade Point Average(SGPA) = -						



Date of Issue: June 3, 2022



K. Venkateswara Rao

CONTROLLER OF EXAMINATIONS

Instructions:

Grading System		
% of Marks Secured in a Subject/Course	Letter Grade	Grade Points
Greater than or equal to 90%	O (Outstanding)	10
80 and less than 90%	A+ (Excellent)	9
70 and less than 80%	A (Very Good)	8
60 and less than 70%	B+ (Good)	7
50 and less than 60%	B (Average)	6
40 and less than 50%	C (Pass)	5
Below 40%	F (FAIL)	0
Absent	AB	0

P: PASS F: FAIL AB: ABSENT

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Ph. : +91 - 8801099936 / 35, 9393808082

Email : principalsdip@sreedattha.ac.in, www.sreedattha.ac.in

B. PHARMACY

Ref. No.

Date: 04/12/2022

11993

Dr. B.Chandra Shekar,

M.Pharm, Ph.D.

Professor & Principal.

To

The Controller of Examination,
JNTU-H,
Kukatpally,
Hyderabad.



04 DEC 2022

Dear Sir/Madam,

Sub: Mr. KANDYANAM PAVAN KUMAR bearing JNTUH No. 19U21R0027-
Grace Marks....Reg

With reference to above subjected, we bring to your kind information that **Mr. KANDYANAM PAVAN KUMAR** bearing JNTUH No. 19U21R0027 is a bonafide student studying B.Pharmacy in our college. He has passed in all the subjects except Cosmetic Science of B.Pharm. III Year I semester. He is applying for the **Grace Marks** based on the result received in February 2022 examination. We are enclosing the marks memo for your kind perusal.

Kindly do the needful and oblige.

Thanking You,

Encl: III/I Marks Memo.



Principal

Sreedattha Institute Of Pharmacy
Sheriguda(V), Ibrahimpatnam(M),
R.R. Dist. Greater Hyderabad - 501510.

